

钱欣博士

Application Specialist

cqian@acs-i.org

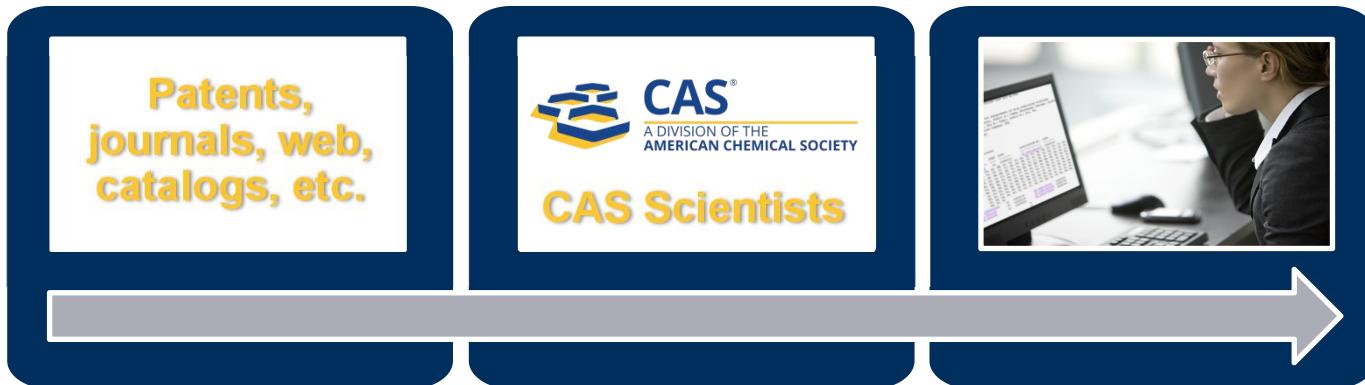
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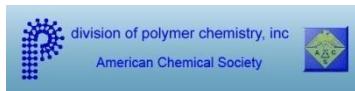


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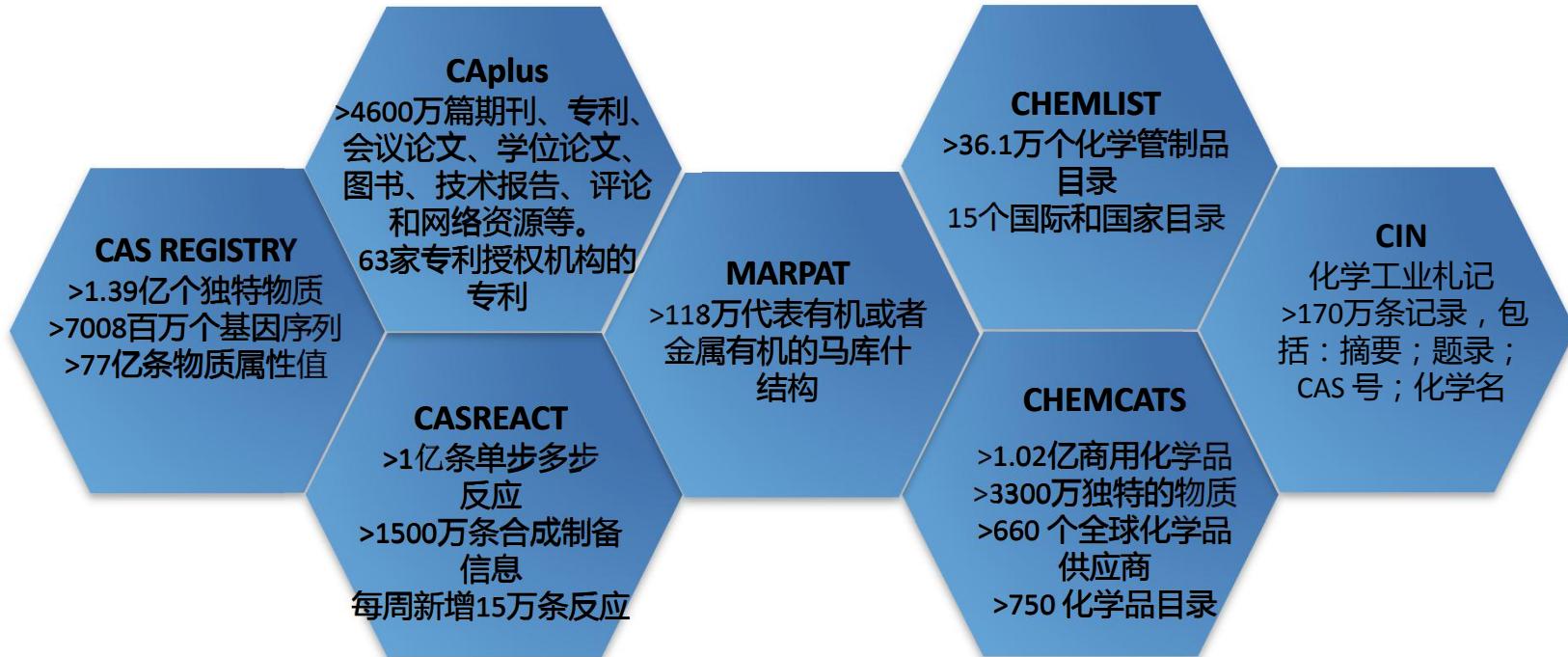


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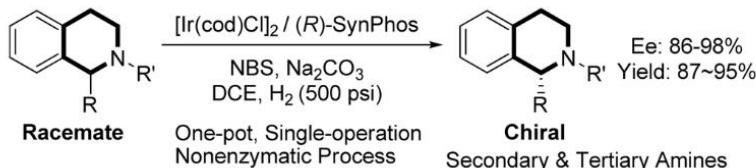


CAS数据库最具价值的内容——人工标引

1. Concise Redox Deracemization of Secondary and Tertiary Amines with a Tetrahydroisoquinoline Core via a Nonenzymatic Process

By: Ji, Yue; Shi, Lei; Chen, Mu-Wang; Feng, Guang-Shou; Zhou, Yong-Gui

A concise deracemization of racemic secondary and tertiary amines with a tetrahydroisoquinoline core has been successfully realized by orchestrating a redox process consisted of N-bromosuccinimide oxidn. and iridium-catalyzed asym. hydrogenation. This compatible redox combination enables one-pot, single-operation deracemization to generate chiral 1-substituted 1,2,3,4-tetrahydroisoquinolines with up to 98% ee in 93% yield, offering a simple and scalable synthetic technique for chiral amines directly from racemic starting materials.



Indexing

Heterocyclic Compounds (One Hetero Atom) (Section27-17)

Concepts

Enantioselective synthesis
Oxidation

Hydrogenation catalysts

stereoselective prepn. of tetrahydroisoquinoline derivs. via iridium-catalyzed deracemization in presence of chiral phosphine ligands

Chiral ligands

stereoselective prepn. of tetrahydroisoquinoline derivs. via iridium-catalyzed deracemization in presence of chiral phosphine ligands

Catalysis Use

Substances

12112-67-3 Dichlorobis(cyclooctadiene)diiridium
76189-55-4
133545-16-1
445467-61-8
503538-68-9 (S)-SynPhos
503538-69-0

stereoselective prepn. of tetrahydroisoquinoline derivs. via iridium-catalyzed deracemization in presence of chiral phosphine ligands

Catalyst use; Uses

QUICK LINKS

0 Tags, 0 Comments

SOURCE

Journal of the American Chemical Society
Volume137
Issue33
Pages10496-10499
Journal; Online Computer File
2015
CODEN:JACSAT
ISSN:0002-7863
DOI:10.1021/jacs.5b06659

COMPANY/ORGANIZATION

State Key Laboratory of Catalysis, Dalian Institute of Chemical Physics
Chinese Academy of Sciences
Dalian, Peop. Rep. China
116023

ACCESSION NUMBER

2015:1340032
CAMSER:221246

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- 如何获得物质的属性信息、谱图信息
- 化合物分析方法检索

- 反应检索

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- 如何进行精确结构反应检索，并获得实验详情
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 - 多氧簇金属的反应信息
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- Property
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- Reaction Structure

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- 检索式：water splitting with solar

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1. Enhanced solar absorption of water splitting composite
By Abed, J.; Almheiri, M.; Alexander, F.; Ra...
From Solar Energy Materials & Solar Cells (2017), Ahead of Print. | Language: English, Database: CAPLUS
We report on the effect of thermal annealing on the microstructure, optical properties and wettability of TiO₂-based **water-splitting** (WS) composite using full **solar** spectrum as source of energy. The WS material used in this study is composed of three layers (SiO₂, Al₂O₃ and TiO₂) on top of which a distribution of Localized Surface Plasmon Resonance structures such as gold are formed to obtain a multilayer composite material. The fabricated samples are then annealed at 450-1100 °C temps. range under atm. conditions. The crystal structure and chem. comprn. are detd. using X-ray Diffractometer,....

2. Photocatalytic water splitting solar-to-hydrogen energy conversion: Perovskite-type hydride XBeH₃ (X = Li or Na) as active photocatalysts
By Reshak, A. H.
From Journal of Catalysis (2017), Ahead of Print. | Language: English, Database: CAPLUS
A highly enhanced photocatalytic hydrogen prodn. system has been achieved, by substitution of Na by Li and moving from cubic to orthorhombic phase in XBeH₃ system. Ab-initio calcns. from first- to second-principles methods were performed to investigate the suitability of the perovskite-type hydride namely; NaBeH₃ and LiBeH₃ in cubic phase and LiBeH₃ in orthorhombic phase to be used as active photocatalysts. We found significant increases in the fundamental energy band gap when we move from NaBeH₃-cubic (0.94 eV) → LiBeH₃-cubic (1.34 eV) → LiBeH₃-orthorhombic (2.44 eV). The obtained energy b...

3. Highly efficient synthesis of hydrogen storage material of formate from bicarbonate and water with general Zn powder
By Song, Jingwen; Yang, Yang; Yao, Guodong; Zhong, Heng; He, Runtian; Jin, Binbin; Jing, Zhenzi; Jin, Fangming
From Industrial & Engineering Chemistry Research (2017), Ahead of Print. | Language: English, Database: CAPLUS
Formate, as an excellent hydrogen-storage material, has recently become increasingly important, because formic acid is low toxic, easy to store and transport, and contains relatively high energy d. In this paper, we give an overview of the recent strategy in the conversion of bicarbonate into formate by **water splitting** with a general metallic Zn powder, which mainly includes: 1) hydrogen

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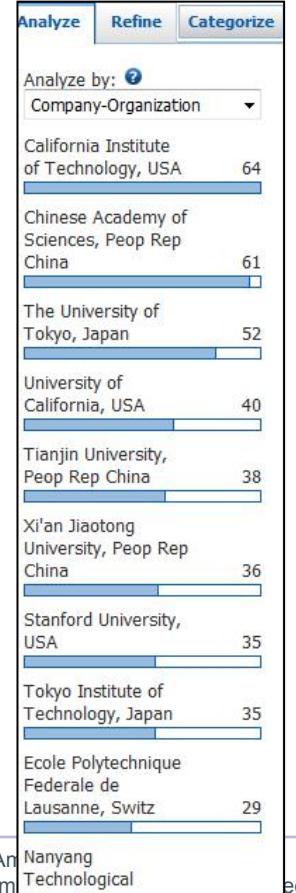
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- Journal Name
- Language
- Publication Year
- Supplementary Terms

主要研究人员

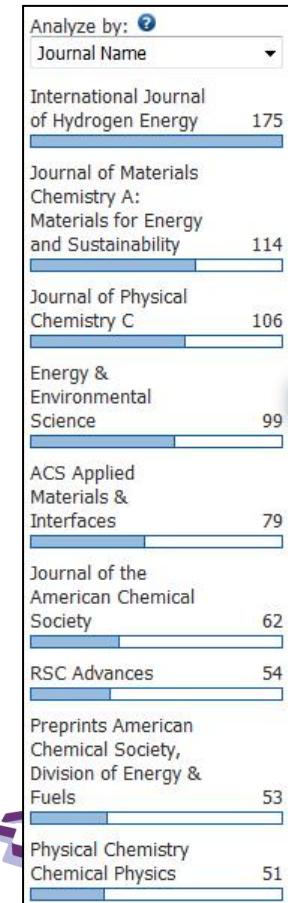


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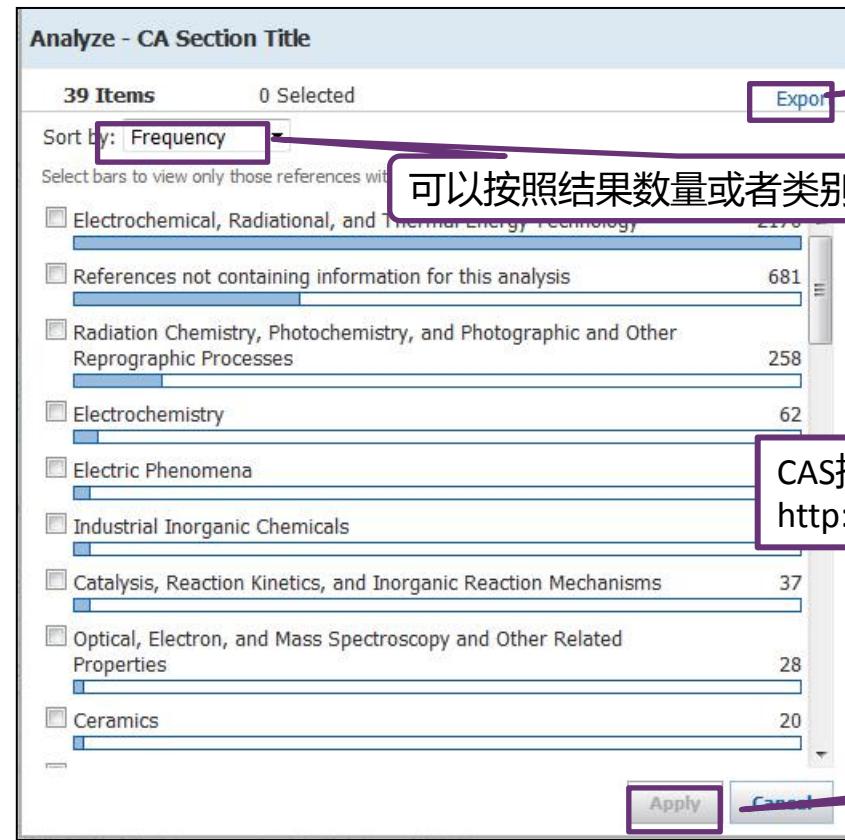
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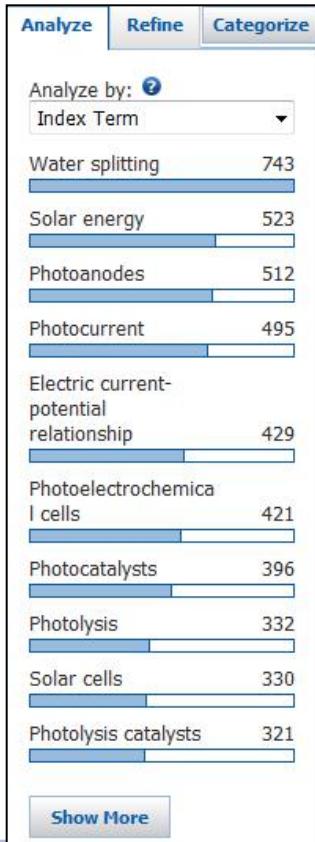
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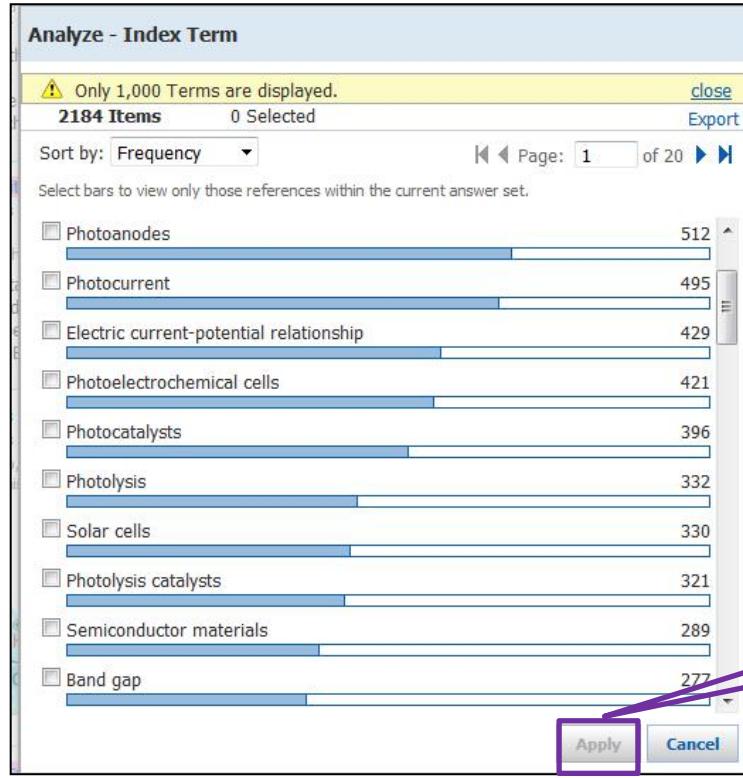
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1. Enhanced solar absorption of gold plasmon assisted TiO₂-based water splitting composite

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By Abed, J.; Almheiri, M.; Alexander, F.; Rajput, N. S.; Viegas, J.; Jouiad, M.
From Solar Energy Materials & Solar Cells (2017), Ahead of Print. | Language: English, Database: CAPLUS

We report on the effect of thermal annealing on the microstructure, optical properties and wettability of TiO₂-based source of energy. The WS material used in this study is composed of three layers (SiO₂, Al₂O₃ and TiO₂) on top of structures such as gold are formed to obtain a multilayer composite material. The fabricated samples are then analyzed. The crystal structure and chem. compn. are detd. using X-ray Diffractometer,...

2. Photocatalytic water splitting solar-to-hydrogen energy conversion: Perovskite-type hydride XBeH₃ (X = Li, Ca)

Quick View Other Sources

By Reshak, A. H.
From Journal of Catalysis (2017), Ahead of Print. | Language: English, Database: CAPLUS

A highly enhanced photocatalytic hydrogen prodn. system has been achieved, by substitution of Na by Li and moving calcns. from first- to second-principles methods were performed to investigate the suitability of the perovskite-type in orthorhombic phase to be used as active photocatalysts. We found significant increases in the fundamental energy LiBeH₃-cubic (1.34 eV) → LiBeH₃-orthorhombic (2.44 eV). The obtained energy b...

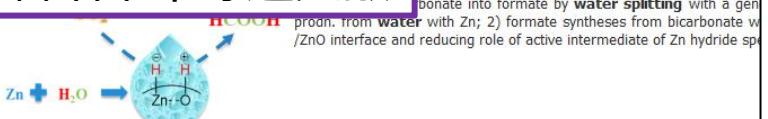
3. Highly efficient synthesis of hydrogen storage material of formate from bicarbonate and water with general

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By Song, Jingwen; Yang, Yang; Yao, Guodong; Zhong, Heng; He, Runtian; Jin, Binbin; Jing, Zhenz; Jin, Fangming
Language: English, Database: CAPLUS

cellent hydrogen-storage material, has recently become a promising candidate for portable energy storage, due to its high capacity, fast transport, and contains relatively high energy density. In this work, we propose a new strategy to synthesize a hydrogen storage material, 1) formate syntheses from bicarbonate with Zn²⁺; 2) formate syntheses from bicarbonate with Zn²⁺ and ZnO interface and reducing role of active intermediate of Zn hydride species.

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Catalysis (49)

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<input type="checkbox"/> Cadmium sulfide 48
<input type="checkbox"/> Bismuth vanadium oxide (BiVO ₄) 47
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307. Efficient water reduction with gallium phosphide nanowires

By: Standing, Anthony; Assali, Simone; Gao, Lu; Verheijen, Marcel A.; van Dam, Dick; Cui, Yingchao; Notten, Peter H. L.; Haverkort, Jos E. M.; Bakkers, Erik P. A. M.

Photoelectrochem. hydrogen prodn. from solar energy and water offers a clean and sustainable fuel option for the future. Planar III/V material systems have shown the highest efficiencies, but are expensive. By moving to the nanowire regime the demand on material quantity is reduced, and new materials can be uncovered, such as wurtzite gallium phosphide, featuring a direct bandgap. This is one of the few materials combining large solar light absorption and (close to) ideal band-edge positions for full water splitting. Here we report the photoelectrochem. redn. of water, on a p-type wurtzite gallium phosphide nanowire photocathode. By modifying geometry to reduce elec. resistance and enhance optical absorption, and modifying the surface with a multistep platinum deposition, high current densities and open circuit potentials were achieved. Our results demonstrate the capabilities of this material, even when used in such low quantities, as in nanowires.

Indexing

Electrochemical, R	Technology (Section52)
Concepts	
Annealing	Band gap
Catalysts	Electric current-potential relationship
Electric resistance	Electrolytes
Evaporation	Fuels
Nanowires	Open circuit potential
Optical absorption	Particle size distribution
Photoelectrochemical reduction	Semiconductor materials
Sol-gel processing	Solar energy
Solutions	pH
efficient water redn. with gallium phosphide nanowires	

Substances
557-20-0 Di-ethyl zinc
1333-74-0 Hydrogen
1445-79-0 Tri-methyl gallium
7440-06-4 Platinum
7601-90-3 Perchloric acid
7803-51-2 Phosphine
9011-14-7 Poly-methyl-methacrylate
12024-21-4 Gallium oxide
12063-98-8 Gallium phosphide
16941-12-1
efficient water redn. with gallium phosphide nanowires
Properties; Technical or engineered material use; Uses

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SOURCE
Nature Communications
Volume6
Pages7824

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4. 文献中重要的物质
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LANGUAGE



PatentPak——专利工作流程解决方案：检索案例：金属储氢材料

- 检索式：hydrogen storage material with metal

23. A hydrogen storage pellet

Quick View PATENTPAK ▾

By Bennington, St
From PCT Int. App

Figure 1

Patent No. WO 2014096865 Kind A1 Language English

Patent Family

US 20140178291	A1	English
KR 2015097788	A	Korean
CN 104884382	A	Chinese
JP 2016502968	T	Japanese

Arthur; Mooring, Lyndsey; Headen, Tom
Language: English, Database: CAPLUS

The **hydrogen storage** pellet is composed of a core of metal particles (<50 µm) to impart structural integrity and protection. The **material** is selected from boron-**hydrogen** compounds, hydride, etc. **Hydrogen** is released, esp. by heating or

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- Li - (CAS RN 7782-89-0) with SMILES: Li - NH₂
- Analyst Markup Location: page 32
- CAS RN 133598-88-6 with SMILES: NH₃ [Cl]⁻²⁺ Mg [Cl]⁻
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PATENTPAK
A CAS SOLUTION

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Analyst Markup Location
page 32

CAS RN 7782-89-0
Li — NH₂
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Analyst Markup Location
page 32

CAS RN 133598-88-6
NH₃
[Cl]⁻²⁺ Mg [Cl]⁻
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et according to any one of claims .
the metal amide or metal ammoniate comprises of
lithium amide-hydride (LiNH₂-LiH), lithium amide,
magnesium chloride ammoniate lithium amide-magnesium
hydride (MgH₂ + LiNH₂), lithium sodium amides,
magnesium amide-lithium hydride (3Mg(NH₂)₂ + 12LiH) or
a mixture of two or more thereof

25

9. The pellet according to any one of claim ,
wherein the metal hydride comprises magnesium hydride,
beryllium hydride, aluminium hydride (alane), lithium
aluminium hydride, sodium aluminium hydride, magnesium
alanate, beryllium alanate, calcium alanate, titanium
alanate lithium-magnesium hexa-alanate or a mixture of
30
two or more thereof.

PatentPak——专利工作流程解决方案

The screenshot shows the PatentPak viewer interface. On the left, there are two sections for 'Key Substances in Patent'. The top section displays the chemical structure of tri-methylamine borane (CAS RN 57808-44-3) with the formula H-B(H)(H)C(H)(H)N. The bottom section displays the chemical structure of ammonium octahydrotriborane (CAS RN 12447-26-6) with the formula [B(H)(H)C(H)(H)N][B(H)(H)C(H)(H)N]2. Both sections include 'Search in SciFinder' and 'View Detail' buttons. The main right area shows patent claims. Claim 5 describes a mixture of methylamine borane, tri-methylamine borane, hydrazine borane, hydrazine bisborane, ethane 1,2-di-amineborane, ammonia triborane, ammonium octahydrotriborane or a mixture of two or more thereof. Claims 6 and 10 describe a pellet comprising ammonia borane, methylamine borane, dimethylamine borane, phenylamine borane, hydrazine borane, hydrazine bisborane, ethane 1,2-di-amineborane, ammonia triborane, ammonium octahydrotriborane or a mixture of two or more thereof. Claim 15 describes a pellet comprising ammonia borane.

也可实现PDF文件与
PatentPak Viewer互动

PatentPak——专利工作流程解决方案

Key Substances in Patent

(12) INTERNATIONAL APPLICATION PUBLISHED
(19) World Intellectual Property

WIPO

CAS Registry Number 57808-44-3

B₃H₉N

Triborane(7), 3-ammine-

Melting Point (Experimental)

Value: 73 °C

Other Names

Triborane(7), ammoniate (6CI)
Triborane(7), monoammoniate
Ammonia triborane
Ammonia-triborane(7)

Chemical structure: B3H9N

(51) International Patent Classification:
C01B 3/00 (2006.01)

(21) International Application Number:
PCT/GB2013/053407

(22) International Filing Date:
20 December 2013 (20.12.2013)

(25) Filing Language:
English

(74) Agent: BOULT WADE TENNANT; Verulam Gardens,
70 Gray's Inn Road, London, Greater London WC1X 8BT
(GB).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IR, IS, JP, KE, KG, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SA, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

可在PatentPak Viewer中直接返回SciFinder进行结构、马库什和反应检索

Chemical structure: B3H9N

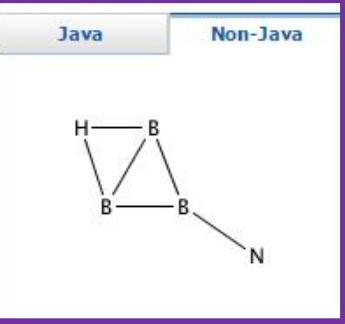
ACS / Proprietary and Confidential / Do Not Distribute

PatentPak——专利工作流程解决方案

SUBSTANCES: CHEMICAL STRUCTURE ?

Structure Editor:

Java Non-Java



Click image to change structure or view detail.

Import CXF

Search

Advanced Search

Search Type:

Exact Structure
 Substructure
 Similarity

Show precision analysis

 ChemDraw®
Launch a SciFinder search

从PatentPak Viewer直接跳转到结构检索界面，同时无需绘制，系统会自动将相应的结构直接导入到绘图面板，点击Search即开始SciFinder检索，拓展新的研究方向

如何使用SciFinder提高科研效率

- 以“光解水”为例，如何利用SciFinder：
 - 获取研究全景
 - 对文献结果进行分析、筛选、及精准定位所需信息
- 物质检索
 - 如何从大量物质结果集中快速定位目标物质
 - 结构式检索；分子式检索（合金，多氧簇金属，MOF化合物）；物质属性检索
 - 如何获得物质的属性信息、谱图信息
反应检索：手性反应检索专题
 - 手性化合物合成方法检索
 - 手性反转反应
 - 手性拆分反应
- 聚合物检索专题
 - 聚合物物质检索
 - 聚合物反应检索



通过文献获得不同研究领域的物质

Explore ▾ Saved Searches ▾ SciPlanner

Research Topic "water splitting with solar" > references (3442) > refine "Journal" (2690) > refine "English" (2570) > refine "CAPLUS" (2458) > refine "2010-" (2148)

REFERENCES ?

Analyze Refine Categorize

Analyze by: Index Term

Water splitting	234
Photoanodes	211
Photocurrent	167
Photocatalysts	158
Photoelectrochemistry	153
Solar energy	152
Electric current-potential relationship	150
Photolysis	131

Get Substances Get Reactions

Sort by: Accession Number

0 of 936 References Selected

Get Substances

Retrieve substances for:

All references
 Selected references

Limit results to:

Adverse Effect, including toxicity
 Analytical Study
 Biological Study
 Combinatorial Study
 Formation, nonpreparative
 Miscellaneous
 Preparation
 Process
 Properties
 Prophetic in Patents
 Reactant or Reagent
 Uses

Get Cancel

In the pursuit of a photosynthetic and efficient water splitting device, efforts have focused on optimizing specific aspects of anode stability, and kinetics among other aspects. Improvement in one aspect can, however, often require fundamental trade-offs to achieve a c.d. of interest for a water splitting device is an esp. important metric. High solar to-hydrogen efficiency is reporting efficiency in a half cell can be misleading. This report invest...

1. Emerging energy applications
Quick View Other Sources
By Late, Dattatray J.; Rout, Chandr From Canadian Chemical Transaction

Atomically thin semiconductors for easy synthesis using various more suitable for nanoelectronics WS₂, MoSe₂, WSe₂, InSe,

2. Understanding photovoltaic cells
Quick View Other Sources
By Scheuermann, Andrew G.; Chidse From ECS Transactions (2015), 69(1)

3. Ultrathin planar hematite film for solar photoelectrochemical water splitting
Quick View Other Sources



光解水中有关用途的物质

SUBSTANCES ? | Get References | Get Reactions | Get Commercial Sources | Tools ▾ | Create Keep Me Posted Alert | Send to SciPlanner

Analyze Refine Sort by: CAS Registry Number ↑ | Display Options | Page: 10 of 18

Analyze by: Substance Role

Uses 877
Properties 867
Process 784
Preparation 778
Reactant or Reagent 695
Formation, Nonpreparative 654
Analytical Study 638
Biological Study 605
Occurrence 589
Miscellaneous 499

Show More

0 of 881 Substances Selected

451. 13494-90-1 (Component: 7697-37-2) ~2448

O=[N+]([O-])OH

• 1/3 Ga

Ga . 3 H N O₃
Nitric acid, gallium salt (3:1)
Regulatory Information
Experimental Properties

452. 13530-65-9 (Component: 7738-94-5) ~1428

Cr(=O)(=O)O(O)O

• Zn

Cr H₂ O₄ . Zn
Chromic acid (H₂CrO₄), zinc salt (1:1)
Key Physical Properties
Regulatory Information
Spectra
Experimental Properties

453. 13587-35-4 ~320

Cu2W

Component Component Ratio

O	4
Cu	1
W	1

Cu . O . W
Copper tungsten oxide (CuWO₄)
Regulatory Information

454. 13595-86-3 ~993

455. 13595-87-4 ~135

456. 13597-19-8 ~274

Component Component Ratio

物质检索结果——Analyze

六种分析选项

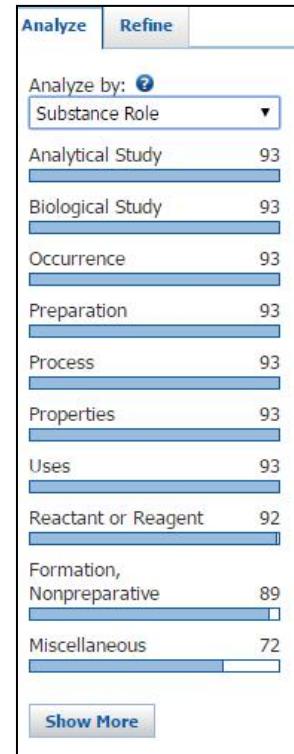
Analyze Refine

Analyze by: ?

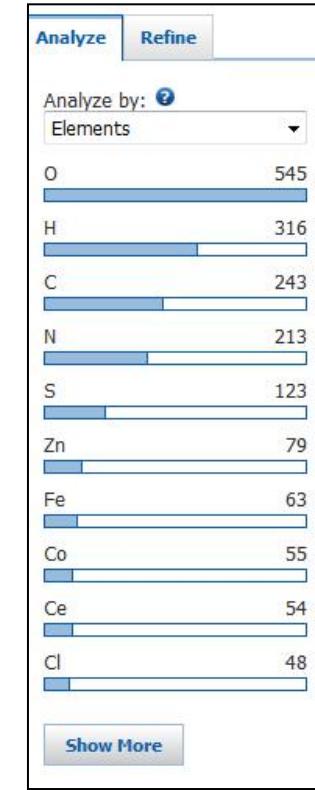
- Bioactivity Indicators
- Bioactivity Indicators
- Commercial Availability
- Elements**
- Reaction Availability
- Substance Role
- Target Indicators

可以按照生物活性，
商业可获得性，元素，
反应可获得性，
物质角色和靶点进行分析

物质角色



元素



物质检索结果——Refine

Analyze Refine

Refine by: ?

- Chemical Structure
- Isotope-Containing
- Metal-Containing
- Commercial Availability
- Property Availability
- Property Value
- Reference Availability
- Atom Attachment

可以按照化学结构，
包含同位素化合物/金属化合物，
商业可获得性，
性质可获得性，
性质数值，文献可获得性
对结果进行限定

Analyze Refine

Refine by: ?

- Chemical Structure
- Isotope-Containing
- Metal-Containing
- Commercial Availability
- Property Availability
- Property Value
- Reference Availability
- Atom Attachment

Retrieve substances with:

- Any property
- Any predicted property
- Any experimental property
- Any selected experimental property

Refine

SUBSTANCES ?

Get Reference

Analyze Refine

Sort by: Re

1. 50 ~21151

C₁₇H₂₁N
8-Azabicyclic
8-methyl-

Key PH
Regulatory
Spectra
Experiment

Select Properties

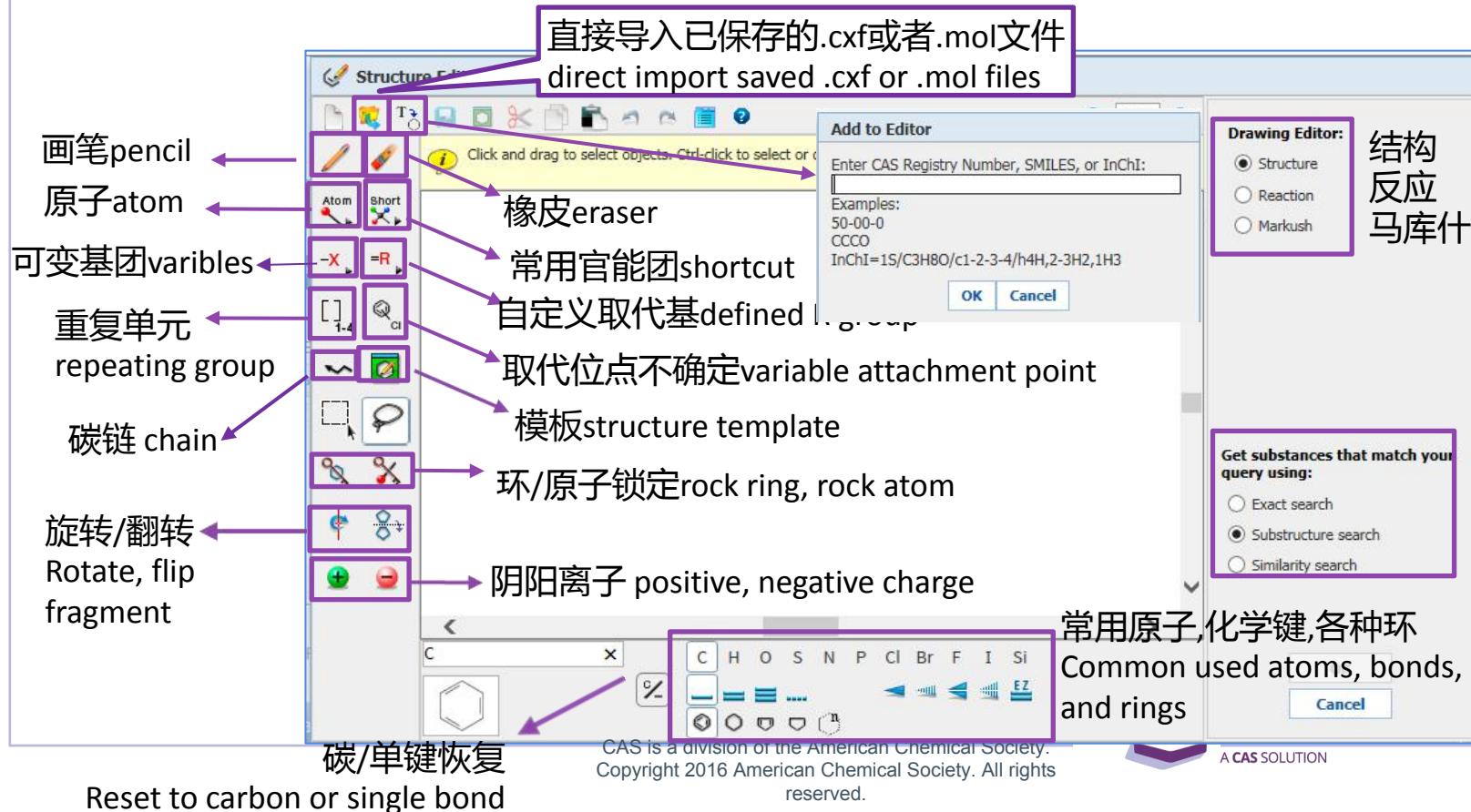
Properties - 0 selected

Experimental

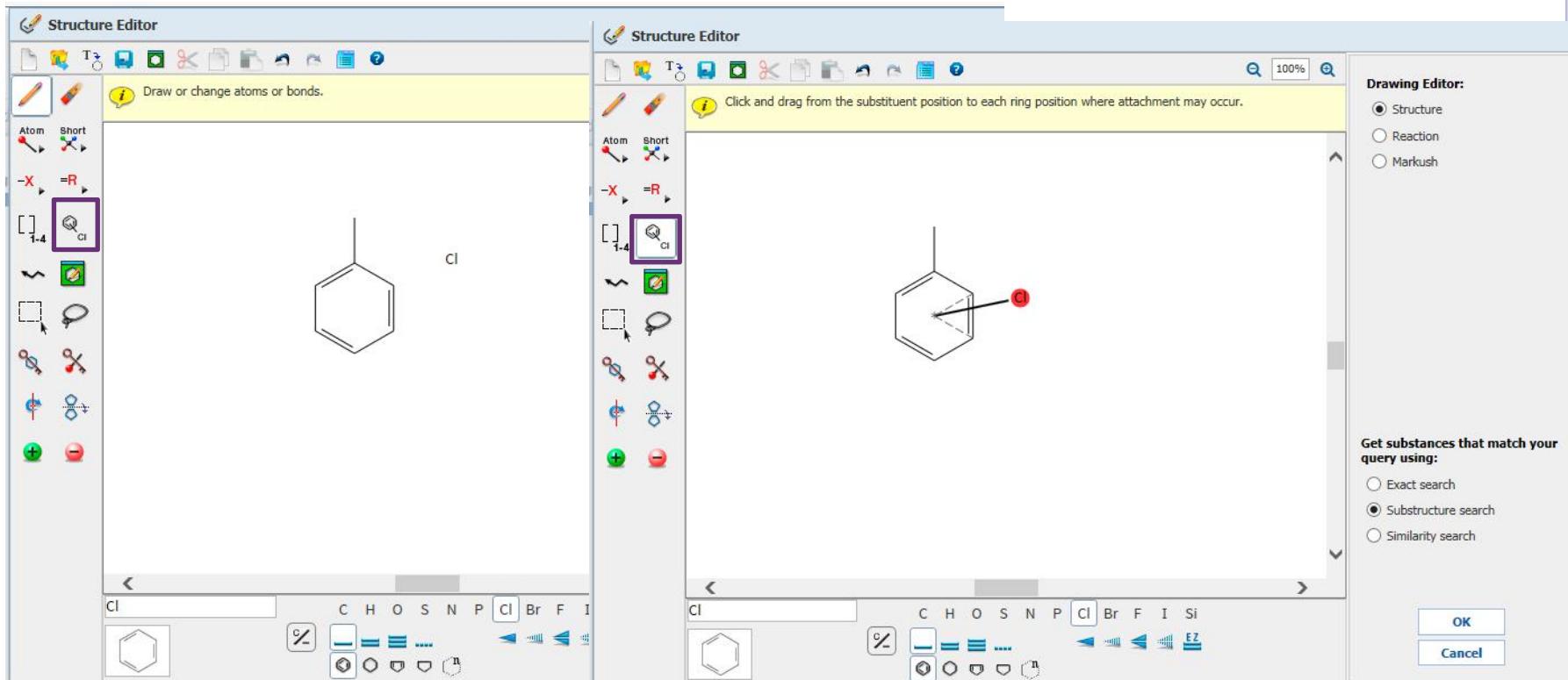
Predicted

- Boiling Point
- Melting Point
- H Acceptors
- H Donors
- Molecular Weight
- logP
- Freely Rotatable Bonds
- Bioconcentration Factor
- Boiling Point
- Density
- Enthalpy of Vaporization
- Flash Point
- H Acceptor/Donor Sum
- Koc
- logD
- Mass Intrinsic Solubility
- Mass Solubility
- Molar Intrinsic Solubility
- Molar Solubility

化学结构检索 : 结构编辑器



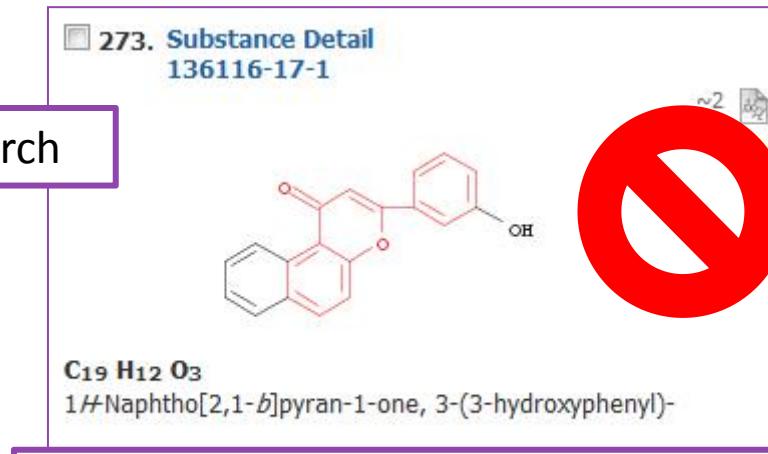
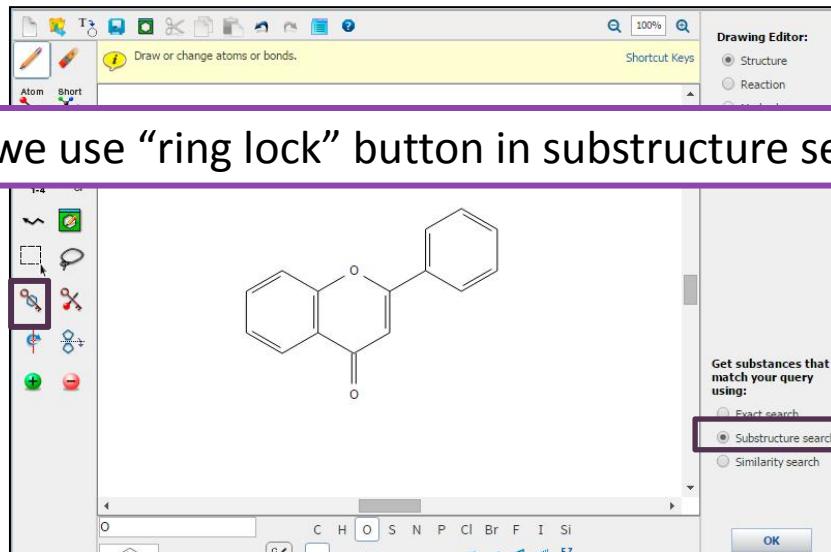
举例：不固定位置取代



举例：环锁定



环锁定，被锁定的结构上不会出现新的环结构

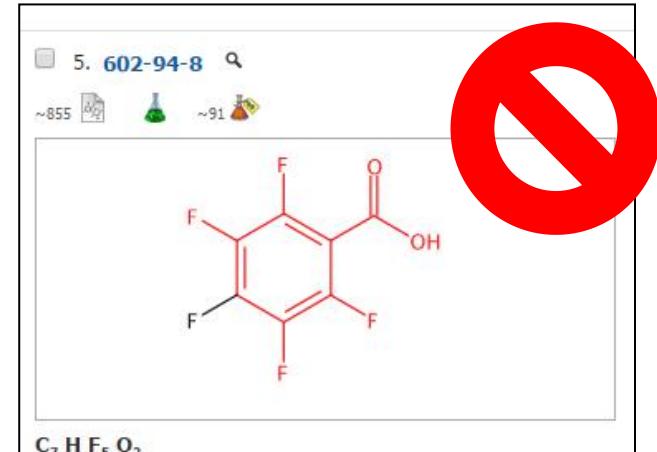
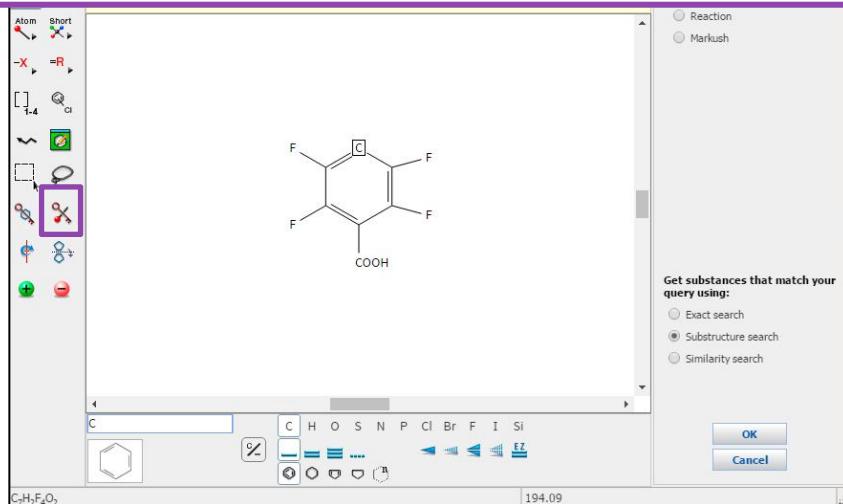


举例：原子锁定



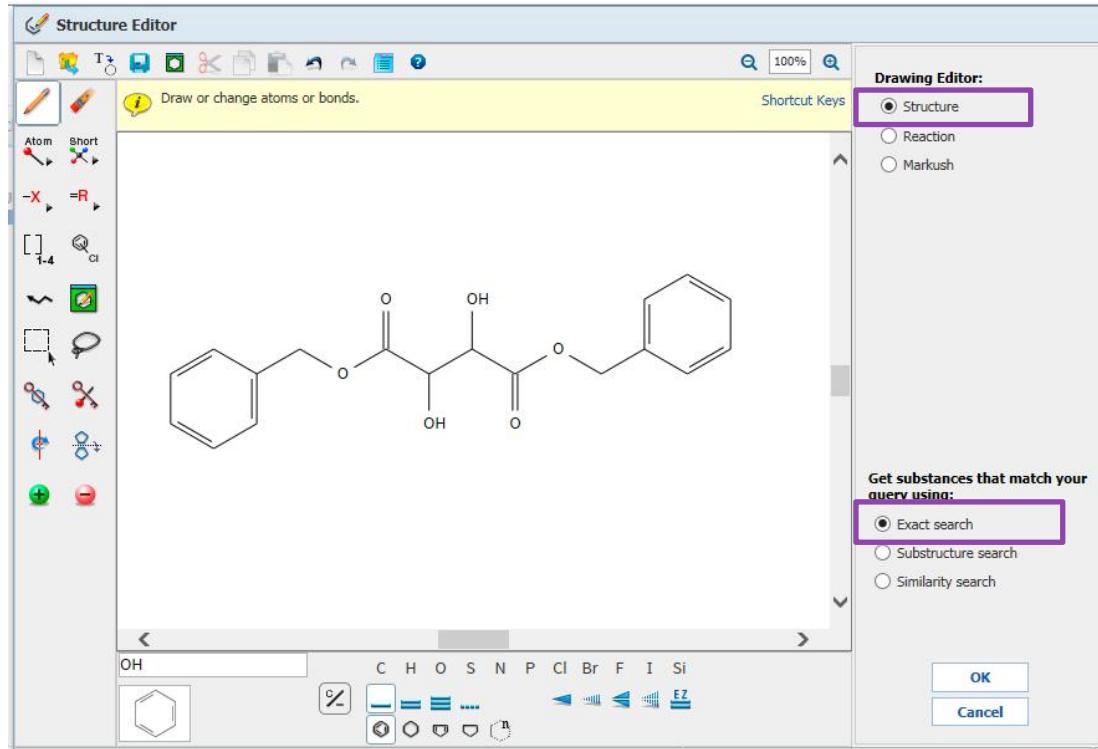
被锁定的原子上不会出现非氢取代

If we use “ring atom” button in substructure search



These type answers will not be retrieved.

化学结构检索: 精确结构检索



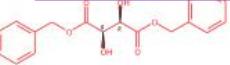
结构式中的原子和环
全部被锁定,不可被取代
All atom and ring are locked,
No substituents

物质结构结果集

0 of 20 Subst

1. 622-00-4   ~105  ~57 

手性化合物 Chiral compound

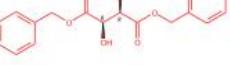


Absolute stereochemistry.,Rotation (+).

C₁₈H₁₈O₆
Butanedioic acid, 2,3-dihydroxy-, (2R,3R)-, 1,4-bis(phenylmethyl) ester

► Key Physical Properties
Experimental Properties

4. 4079-56-5   ~5  ~1 

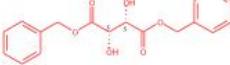


Relative stereochemistry.

C₁₈H₁₈O₆
Butanedioic acid, 2,3-dihydroxy-, 1,4-bis(phenylmethyl) ester, (2R,3S)--rel

精确结构检索获得被检索结构的盐，
混合物，配合物，聚合物等，
被检结构不能被取代

2. 4136-22-5   ~46 

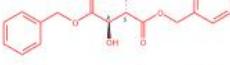


Absolute stereochemistry.,Rotation (-).

C₁₈H₁₈O₆
Butanedioic acid, 2,3-dihydroxy-, 1,4-bis(phenylmethyl) ester, (2S,3S)-

► Key Physical Properties

5. 4079-57-6   ~5  ~1 

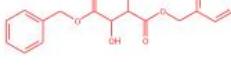


Relative stereochemistry.

C₁₈H₁₈O₆
Butanedioic acid, 2,3-dihydroxy-, 1,4-bis(phenylmethyl) ester, (2R,3S)-rel

Key Physical Properties

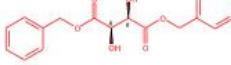
3. 93993-87-4   ~9  ~7 



C₁₈H₁₈O₆
Butanedioic acid, 2,3-dihydroxy-, 1,4-bis(phenylmethyl) ester

► Key Physical Properties

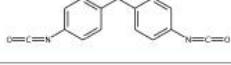
6. 89960-73-6   ~2  



Absolute stereochemistry.,Rotation (+).

C₁₈H₁₈O₆
Butanedioic acid, 2,3-dihydroxy-, (2R,3R)-, 1,4-bis(phenylmethyl) ester

101-68-8   C₁₅H₁₀N₂O₂ 



非手性化合物 Non-chiral compound

聚合物单体的组合物 Monomer of polymer

高级检索——提前限定

Search

Advanced Search Always Show

Characteristics Single component
 Commercially available
 Included in references

Classes Alloys
 Coordination compounds
 Incompletely defined
 Mixtures
 Polymers
 Organics, and others not listed

Studies Analytical
 Biological
 Preparation
 Reactant or reagent



0 of 7 Substances Selected

1. 622-00-4    

O=C(O[C@H](C(=O)OCc1ccccc1)[C@H](O)[C@H](O)[C@H]1CCCCC1)C(=O)OCc2ccccc2
Absolute stereochemistry.,Rotation (+).

C18H18O6
Butanedioic acid, 2,3-dihydroxy-, (2R,3R)-, 1,4-bis(phenylmethyl) ester

Key Physical Properties
Experimental Properties

4. 4079-56-5   

O=C(O[C@H](C(=O)OCc1ccccc1)[C@H](O)[C@H](O)[C@H]1CCCCC1)C(=O)OCc2ccccc2
Relative stereochemistry.

C18H18O6
Butanedioic acid, 2,3-dihydroxy-, 1,4-bis(phenylmethyl) ester,
(2S,3R)-rel

Key Physical Properties

3. 93993-87-4    

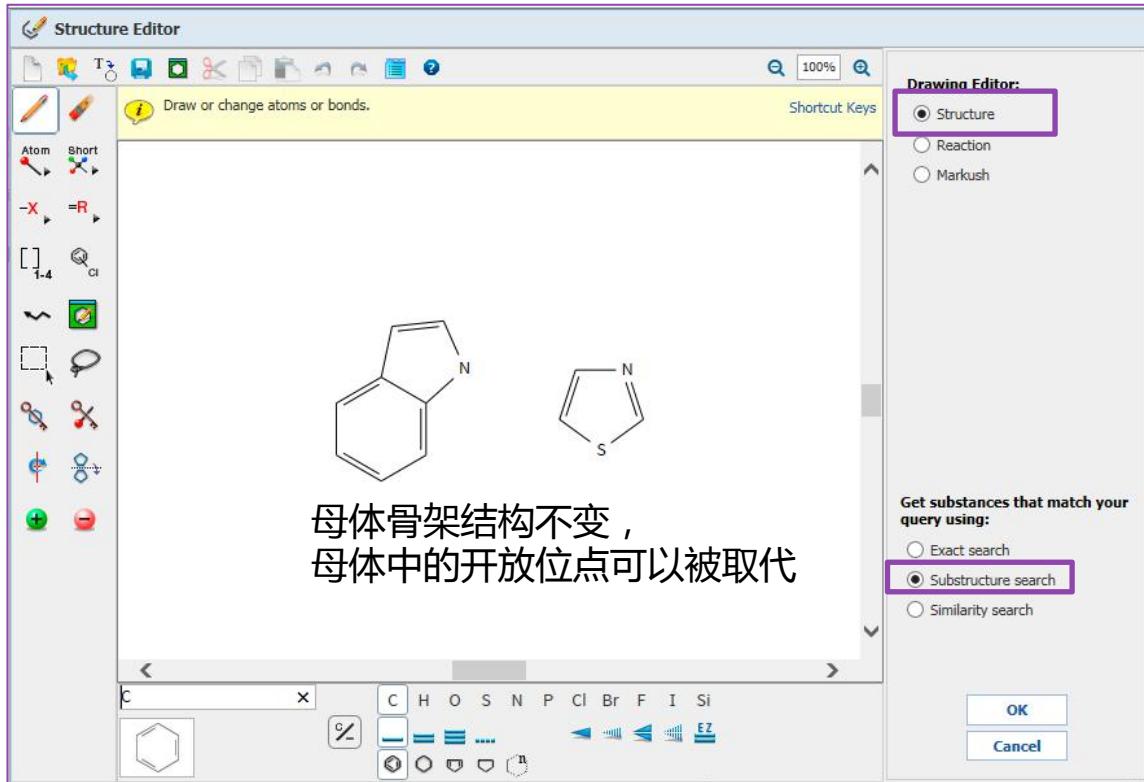
O=C(O[C@H](C(=O)OCc1ccccc1)[C@H](O)[C@H](O)[C@H]1CCCCC1)C(=O)OCc2ccccc2
Absolute stereochemistry.

C18H18O6
Butanedioic acid, 2,3-dihydroxy-, 1,4-bis(phenylmethyl) ester,
(2S,3R)-

Key Physical Properties

去除混合物，组合物，配合物，盐
Exclude mixture, polymer, complex, salts

化学结构检索：亚结构检索



The screenshot shows the SciFinder Advanced Search interface. It includes sections for 'Characteristics' (with 'Single component' checked), 'Classes' (with 'Organics, and others not listed' checked), and 'Studies' (with 'Analytical', 'Biological', 'Preparation', and 'Reactant or reagent' options). The 'Advanced Search' button is highlighted.

将两个结构片段限定在
同一物质中

物质结果集

Chemical Structure substructure with limiters > substances (49240)

SUBSTANCES

Analyze Refine

Sample Analysis: Substance Role

Preparation ≥ 6761

Biological Study ≥ 5629

Uses ≥ 4822

Prophetic in Patents ≥ 1649

Reactant or Reagent ≥ 1241

Properties ≥ 369

Process ≥ 49

Combinatorial Study ≥ 25

Analytical Study ≥ 17

Occurrence ≥ 7

Show More

Get References

Get Reactions

Get Commercial Sources

Tools

Sort by: CAS Registry Number

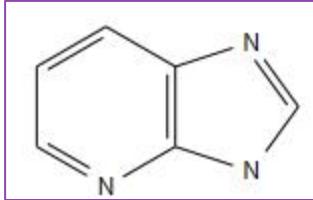
0 of 49240 Substances Selected

Display Options

Page: 1 of 3283

<p>□ 1. 1997357-70-6 </p> <p></p> <p>C₂₀H₁₈N₄O₃S₂ Acetamide, N-[4-[3-[(1-methyl-1H-indol-4-yl)sulfonyl]amino]phenyl]-2-thiazolyl]-</p> <p>► Key Physical Properties</p>	<p>□ 2. 1997357-68-2 </p> <p></p> <p>C₂₀H₁₈N₄O₃S₂ Acetamide, N-[4-[3-[(1-methyl-1H-indol-4-yl)sulfonyl]amino]phenyl]-2-thiazolyl]-</p> <p>► Key Physical Properties</p>	<p>□ 3. 1995871-38-9 </p> <p></p> <p>C₃₀H₃₂N₅S Pyridinium, 4-[2-[6-[[2-(2-benzothiazolyl)hydrazinylidene]methyl]-9-ethyl-9H-carbazol-3-yl]ethenyl]-1-methyl-</p>
<p>□ 4. 1995847-80-7 </p> <p></p> <p>Absolute stereochemistry.</p> <p>C₄₀H₅₈N₆O₇S 1H-Indole-3-pentanoic acid, γ-[[[2-[(1R,3R)-1-(acetoxy)-4-</p>	<p>□ 5. 1995846-52-0 </p> <p></p> <p>Absolute stereochemistry.</p> <p>C₄₀H₅₈N₆O₇S 1H-Indole-3-pentanoic acid, γ-[[[2-[(1R,3R)-1-(acetoxy)-4-</p>	<p>□ 6. 1995844-98-8 </p> <p></p> <p>Absolute stereochemistry.</p>

化学结构检索: 相似结构检索



0 of 7 Similarity Candidates Selected

Similarity Range	Substances
≥ 99 (most similar)	26
95-98	0
90-94	0
85-89	8
80-84	23
75-79	179
70-74	199
65-69	997
0-64 (least similar)	2321

Get Substances

相似度越高，结构越相似

Purin

母体结构被拆分
成分子片段

Score: 81
 14. [128710-59-8](#)

~0

C₆H₅N₃
3*H*-Imidazo[4,5-*b*]pyridine

Score: 73
 36. [42341-33-3](#)

~0

C₄H₃BN₄
1*H*-Imidazo[4,5-*d*]-1,3,2-diazaborine (9CI)

物质检索——理化性质检索：寻找电阻率大于125ohm的含铁物质

The screenshot shows the SciFinder interface with the following details:

- Top Navigation:** CAS Solutions, SCI-FINDER, A CAS SOLUTION.
- Header:** Explore ▾, Saved Searches ▾, SciPlanner.
- Left Sidebar:**
 - REFERENCES**: Research Topic, Author Name, Company Name, Document Identifier, Journal, Patent, Tags.
 - SUBSTANCES**: Chemical Structure, Markush, Molecular Formula, Property, Substance Identifier.
 - REACTIONS**: Reaction Structure.
- Main Content:** SUBSTANCES: PROPERTY ?
 - Property Type:** Experimental (radio button selected).
 - Select Property...** dropdown menu:
 - Boiling Point (°C)
 - Density (g/cm³)
 - Electric Conductance (S)
 - Electric Conductivity (S/cm)
 - Electric Resistance (ohm)
 - Electric Resistivity (ohm*cm)
 - Glass Transition Temp. (°C)
 - Magnetic Moment (μ B)
 - Median Lethal Dose (LD50) (mg/kg)
 - Melting Point (°C)
 - Optical Rotatory Power (degrees)
 - Refractive Index
 - Tensile Strength (MPa)
 - Examples:** Examples: 44, 25-35, >125

物质检索——理化性质性质检索

The screenshot shows the SciFinder search interface. On the left, there are navigation links for 'REFERENCES', 'SUBSTANCES', and 'REACTIONS'. The 'SUBSTANCES' section is currently selected. In the center, a dropdown menu titled 'Select Property...' lists various chemical properties. The 'Molecular Weight' option is highlighted with a blue selection bar. Below the dropdown, there is a search bar containing the placeholder text 'Examples: 44, 25-35, >125'. At the bottom of the search area is a large blue 'Search' button.

- Select Property...
- Bioconcentration Factor
- Boiling Point (°C)
- Density (g/cm³)
- Enthalpy of Vaporization (kJ/mol)
- Flash Point (°C)
- Freely Rotatable Bonds
- H Donor/Acceptor sum
- H Acceptors
- H Donors
- Koc
- logD
- logP
- Mass Intrinsic Solubility (g/L)
- Mass Solubility (g/L)
- Molar Intrinsic Solubility (mol/L)
- Molar Solubility (mol/L)
- Molar Volume (cm³/mol)
- Molecular Weight**
- pKa
- Select Property...

SciFinder物质检索结果

SUBSTANCES ?

Get References Get Reactions Get Commercial Sources Tools ▾

Create Keep Me Posted Alert Send to SciPlanner

Analyze Refine Sort by: CAS Registry Number ▾

Display Options

Refine by: ?

- Chemical Structure
- Isotope-Containing
- Metal-Containing
- Commercial Availability
- Property Availability
- Property Value
- Reference Availability
- Atom Attachment

0 of 137 Substances Selected

1. 1360099-47-3

~189

Component	Component Ratio
Te	x
Cd	x
Hg	x

Cd . Hg . Te
Cadmium mercury telluride
Experimental Properties

2. 1262894-47-2

~2

Double bond geometry as shown., Relative stereochemistry.

C₂₅ H₂₀ N₄ O
2-Azetidinone, 1-phenyl-4-[(1*E*)-2-phenylethenyl]-3-(4-phenyl-1*H*-1,2,3-triazol-1-yl)-, (3*R*,4*S*)-*rel*

Key Physical Properties
Experimental Properties

3. 1160936-40-2

~1

Component	Component Ratio
Te	1.8
Se	0.2
In	0.1
Cu	0.1
Cd	1.8

Cd . Cu . In . Se . Te
Cadmium copper indium selenide telluride
(Cd_{1.8}Cu_{0.1}In_{0.1}Se_{0.2}Te_{1.8})
Experimental Properties

4. 1160936-38-8

~1

5. 1160936-37-7

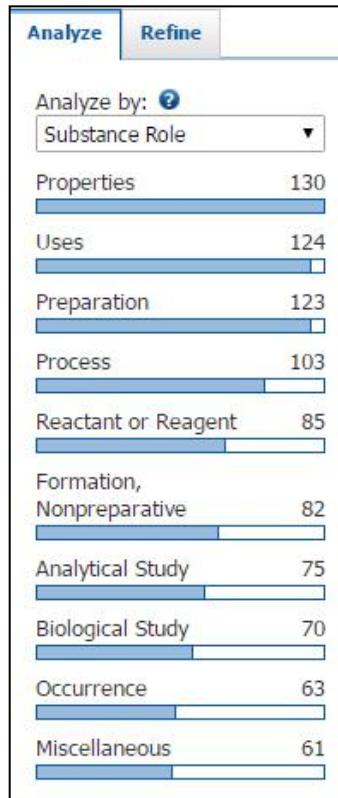
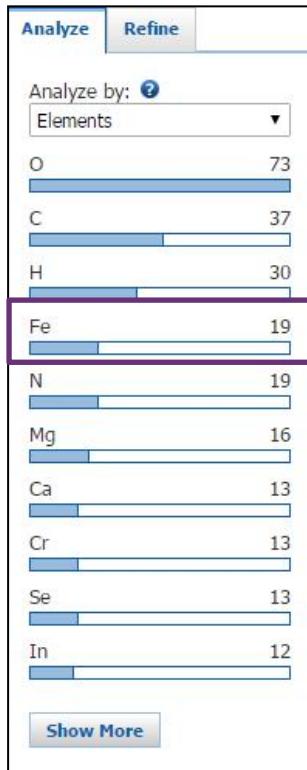
~1

6. 1034343-98-0

~86194

~48

通过分析/限定工具筛选结果



Analyze Refine

Refine by:

- Chemical Structure
- Isotope-Containing
- Metal-Containing
- Commercial Availability
- Property Availability
- Property Value
- Reference Availability
- Atom Attachment

Structure Editor:

Java Non-Java

Click to Edit

Search type: Exact Structure

Only retrieve substances that:

Only retrieve substances that:

- Have references
- Are commercially available
- Are a single component
- Are in specific substance classes
 - Alloys
 - Coordination compounds
 - Incompletely defined
 - Mixtures
 - Polymers
 - Organics, and others not listed
- Are in specific types of studies

Refine

提纲

- SciFinder物质检索
 - 通过文献获得物质
 - 物质检索结果的处理：分析和限定
 - 物质信息详情
 - 理化性质检索
 - 分子式检索
 - 物质标识符检索

物质检索——分子式检索

REFERENCES

- Research Topic
- Author Name
- Company Name
- Document Identifier
- Journal
- Patent
- Tags

SUBSTANCES

- Chemical Structure
- Markush
- Molecular Formula**
- Property
- Substance Identifier

REACTIONS

- Read

SUBSTANCES: MOLECULAR FORMULA

C₁₂H₂₆O₄S.Na

Examples:
H₄SiO₄
(C₃H₆O.C₂H₄O)_x

Search

无机金属盐：金属离子和阴离子间用点（.）分开

40. 151-21-3

(Component: 151-41-7)

~79363 ~283

The chemical structure shows a central sulfur atom bonded to two oxygen atoms (forming two sulfate groups) and one methyl group. It is also bonded to a long-chain dodecyl group (-CH₂)₁₁-CH₃. Below the structure is a sodium ion (• Na).

C₁₂H₂₆O₄S.Na
Sulfuric acid monododecyl ester sodium salt (1:1)

Key Physical Properties

- Regulatory Information
- Spectra
- Experimental Properties

分子式输入需要遵守Hill排序规则:不含碳化合物,按元素符号的字母顺序排列;分子式为含碳化合物时,则“C”在前;如有氢则紧随其后,其它元素符号按字母顺序排在氢的后面

SCI-FINDER®
A CAS SOLUTION

ACS / Proprietary and Confidential / Do Not Distribute

43

合金的检索：钴铁锰合金

Molecular Formula "Co . Fe . Mn" > substances (249)

REFERENCES

- Research Topic
- Author Name
- Company Name
- Document Identifier
- Journal
- Patent
- Tags

SUBSTANCES

- Chemical Structure
- Markush
- Molecular Formula**
- Property
- Substance Identifier

REACTIONS

- Reaction Structure

SUBSTANCES: MOLECULAR FORMULA ?

Co.Fe.Mn
Examples:
H4SiO4
(C3H6O.C2H4O)x

Sort by:

- CAS Registry Number
- CAS Registry Number
- Number of References
- Number of Commercial Sources
- Molecular Weight
- Molecular Formula**

Tools:

- Create Keep Me Posted Alert
- Send to SciPlanner

Display Options | Page: 1 of 5

用“.”将不同组分隔开
使用排序功能（可按照第一个组分的成分排序提高检索效率）

Component	Component Percent
Mn	69
Fe	26
Co	4.3

Co . Fe . Mn
Manganese alloy, base, Mn 69,Fe 26,Co 4.3

Component	Component Percent
Mn	70
Fe	28
Co	2.1

2. 1632299-35-4

Component	Component Percent
Mn	70
Fe	28
Co	2.1

3. 1612139-04-4

Component	Component Percent
Co	35
Fe	33
Mn	32

Co . Fe . Mn
Cobalt alloy, base, Co 35,Fe 33,Mn 32

Component	Component Percent
Fe	62 - 79
Mn	20 - 32
Co	1.1 - 9.5

4. 1585186-25-9

Component	Component Percent
Co	34
Fe	34
Mn	32

5. 1446711-14-3

Component	Component Percent
Fe	62 - 79
Mn	20 - 32
Co	1.1 - 9.5

6. 1383688-58-1

Component	Component Ratio
Co	30
Mn	3
Fe	70

多氧簇金属：七钼酸铵

CAS Registry Number 12027-67-7

(Component: 12274-10-1)



$\text{H}_4\text{N} \cdot \frac{1}{6} \text{Mo}_7\text{O}_{24}$

Molybdate ($\text{Mo}_7\text{O}_{24}^{5-}$), ammonium (1:6)

Coordination Compound

Other Names

Ammonium molybdate(VI) ($(\text{NH}_4)_6\text{Mo}_7\text{O}_{24}$) (6CI)

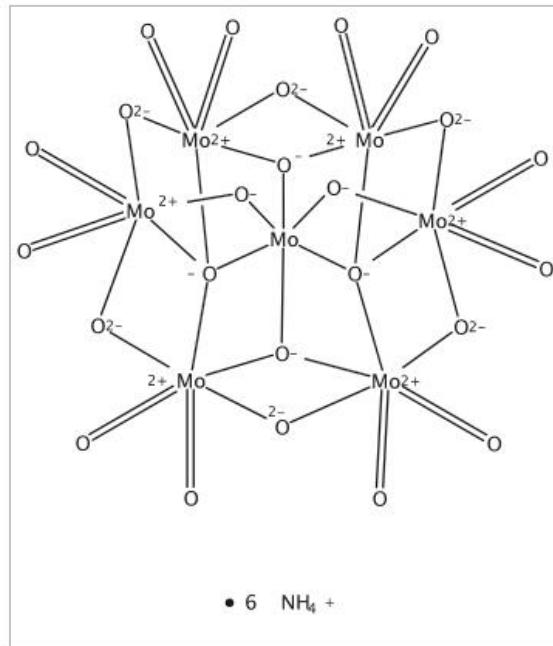
Molybdate ($\text{Mo}_7\text{O}_{24}^{5-}$), hexaammonium (9CI)

Molybdic acid ($\text{H}_6\text{Mo}_7\text{O}_{24}$), hexaammonium salt (8CI)

Ammonium heptamolybdate

Ammonium heptamolybdate ($(\text{NH}_4)_6\text{Mo}_7\text{O}_{24}$)

[View more...](#)



$\text{H}_4\text{N} \cdot \frac{1}{6} \text{Mo}_7\text{O}_{24}$

如何查找MOF化合物，以Cu-BTC为例

Substance Identifier "222404-02-6" > substances (1) > 222404-02-6

SUBSTANCE DETAIL

[Return](#)

CAS Registry Number 222404-02-6

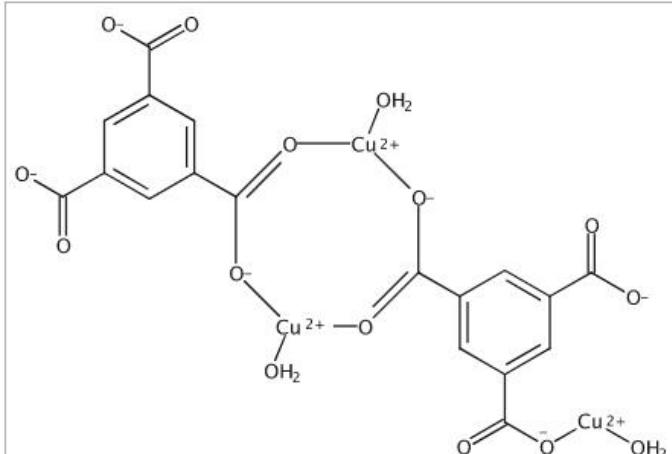
~1,274

C₁₈H₁₂Cu₃O₁₅

Copper, triqua[μ-[1,3,5-benzenetricarboxylato(3)-κO⁴:κO¹]][μ₃-[1,3,5-benzenetricarboxylato(3)-κO⁴:κO³:κO¹]]tricopper
Coordination Compound

Density (Experimental)
Value: 0.879 g/cm³

Other Names
Copper, triqua[μ-[1,3,5-benzenetricarboxylato(3)-κO⁴:κO¹]][μ₃-[1,3,5-benzenetricarboxylato(3)-κO⁴:κO³:κO¹]]tricopper (9CI)
Triqua[μ-[1,3,5-benzenetricarboxylato(3)-κO⁴:κO³:κO¹]][μ₃-[1,3,5-benzenetricarboxylato(3)-κO⁴:κO³:κO¹]]tricopper
Basolite C 300
Cu-BTC
Cu₃BTC₂
[View more...](#)



方法1：若已知物质的CAS号码，物质名称，使用Substance Identifier

Explore ▾ Saved Searches ▾ SciPlanner

Substance Identifier "222404-02-6" > substances (1) > 222404-02-6

REFERENCES

- Research Topic
- Author Name
- Company Name
- Document Identifier
- Journal
- Patent
- Tags

SUBSTANCES

- Chemical Structure
- Markush
- Molecular Formula
- Property
- Substance Identifier**

REACTIONS

- Reaction Structure

SUBSTANCES: SUBSTANCE IDENTIFIER ?

222404-02-6

Enter one per line.
Examples:
50-00-0
999815
Acetaminophen

Search

Explore ▾ Saved Searches ▾ SciPlanner

Substance Identifier "Cu-BTC" > substances (1)

REFERENCES

- Research Topic
- Author Name
- Company Name
- Document Identifier
- Journal
- Patent
- Tags

SUBSTANCES

- Chemical Structure
- Markush
- Molecular Formula
- Property
- Substance Identifier**

REACTIONS

- Reaction Structure

SUBSTANCES: SUBSTANCE IDENTIFIER ?

Cu-BTC

Enter one per line.
Examples:
50-00-0
999815
Acetaminophen

Search

方法2：已知MOF化合物单元分子式，可以使用分子式检索

Explore ▾ Saved Searches ▾ SciPlanner

Chemical Structure exact with limiters > substances (7)

REFERENCES

- Research Topic
- Author Name
- Company Name
- Document Identifier
- Journal
- Patent
- Tags

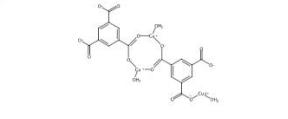
SUBSTANCES

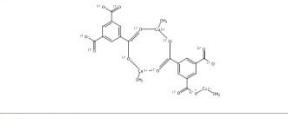
- Chemical Structure
- Markush
- Molecular Formula**
- Property
- Substance Identifier

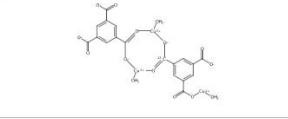
REACTIONS

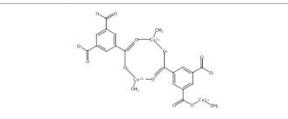
- Reaction Structure

0 of 4 Substances Selected

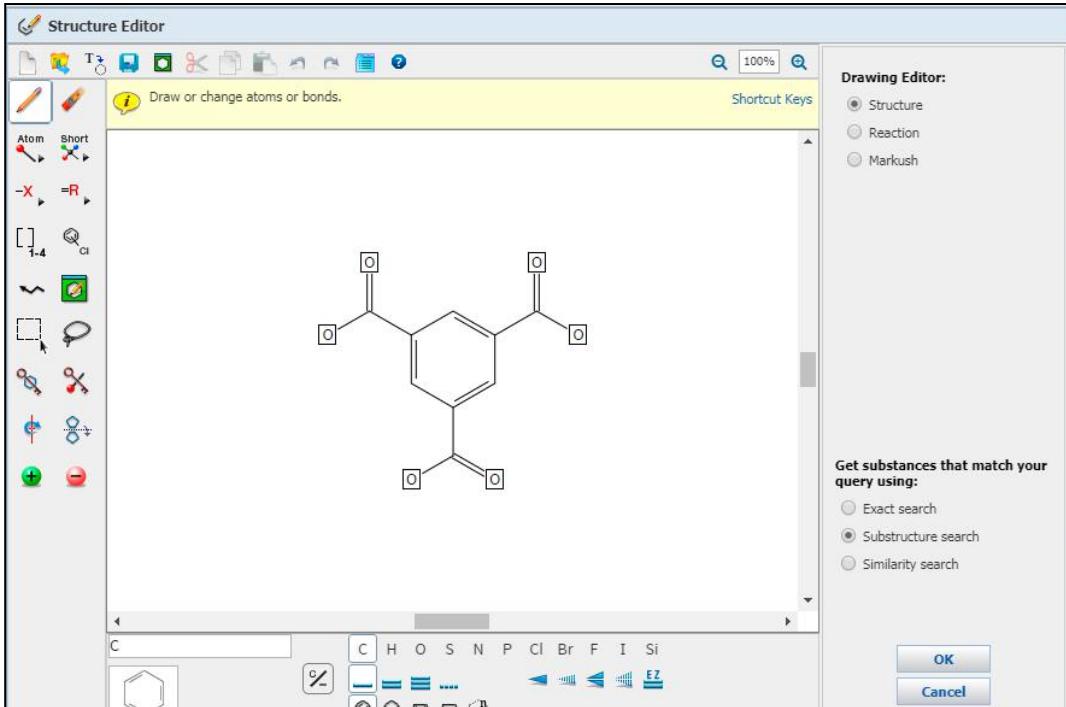
1. 2019181-18-9 
C18H12Cu3O15
INDEX NAME NOT YET ASSIGNED

2. 1685249-53-9 
C18H12Cu3O15
INDEX NAME NOT YET ASSIGNED

3. 1416961-85-7 
C18H12Cu3O15
Copper triaquatri-[1,3,5-benzenetricarboxylato(3-)xO²⁻xO¹⁻][μ₃-[1,3,5-benzenetricarboxylato(3-)xO²⁻xO¹⁻]]
tri-

4. 222404-02-6 
C18H12Cu3O15

方法3：已知配合物配体，可以先结构检索配体，然后获得相关配合物



Search

Advanced Search Always Show

Characteristics

- Single component
- Commercially available
- Included in references

Classes

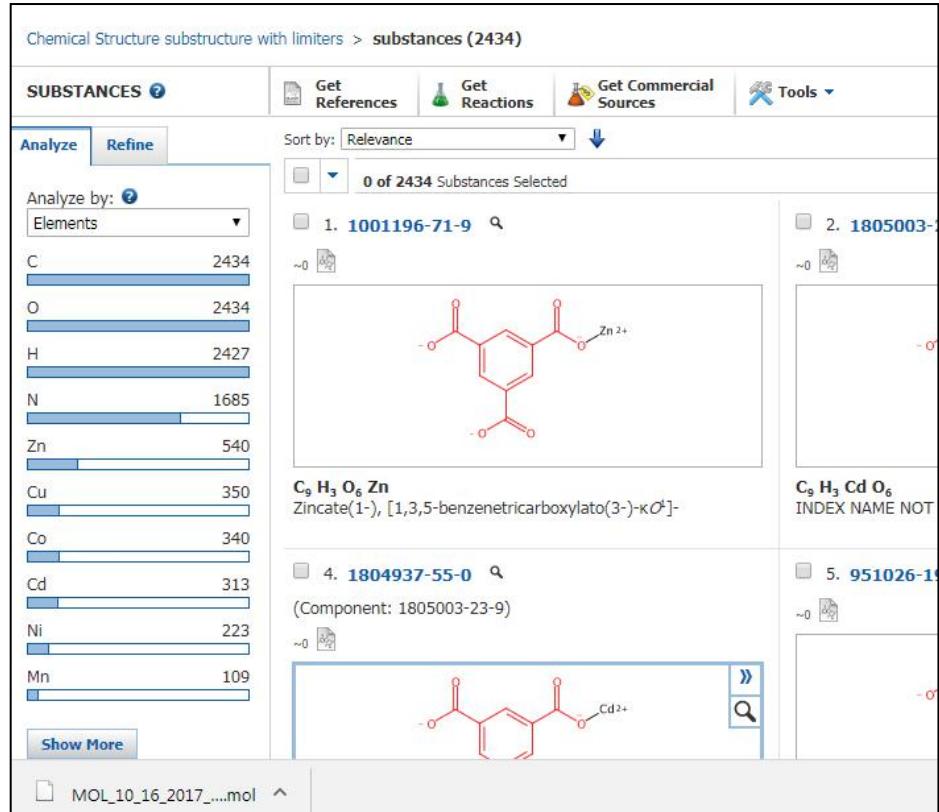
- Alloys
- Coordination compounds
- Incompletely defined
- Mixtures
- Polymers
- Organics, and others not listed

Studies

- Analytical
- Biological
- Preparation
- Reactant or reagent

注：此种方法可能结果噪音较多

在结果集中通过Analyze by Elements , 选择Cu



结果会有噪音，可以通过结构等进一步进行筛选

Explore ▾ Saved Searches ▾ SciPlanner

Chemical Structure substructure with limiters > substances (2434) > keep analysis "Elements" (350)

SUBSTANCES ?

Analyze Refine

Analyze by: Elements

C 350

Cu 350

H 350

O 350

N 288

Cl 16

S 15

Mn 6

Ca 5

Sort by: Number of References

Relevance

CAS Registry Number

Number of References

1. Number of Commercial Sources

Molecular Weight

Molecular Formula

~1274

2. 222403-98-7

(Component: 222404-02-6)

~38

*C(=O)c1ccc(cc1)[C@H](OCC(C)C(=O)O)C[C@H]2[C@@H](OCC(C)C(=O)O)C(=O)c3cc(OCC(C)C(=O)O)cc23

C18H12Cu3O15

Copper, triqua[μ-[1,3,5-benzenetricarboxylato(3-)·K⁺·O²⁻]]
[μ₃-[1,3,5-benzenetricarboxylato(3-)·K⁺·O²⁻]]tri-

► Key Physical Properties

Experimental Properties

C18H12Cu3O15.xH2O

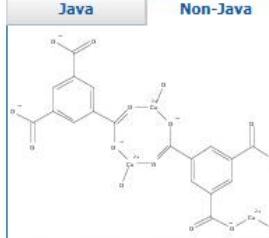
Copper, triqua[μ-[1,3,5-benzenetricarboxylato(3-)·K⁺·O²⁻]]
[μ₃-[1,3,5-benzenetricarboxylato(3-)·K⁺·O²⁻]]tri-,
hydrate (1:?)

• x H₂O

方法4：已知MOF化合物单元结构式，可以绘制完整结构进行检索

Structure Editor:

Java Non-Java



Click image to change structure or view detail.

Import CXF

Search

Advanced Search Always Show

Characteristics

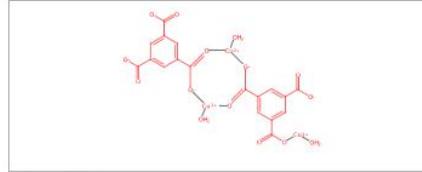
- Single component
- Commercially available
- Included in references

Classes

- Alloys
- Coordination compounds

0 of 7 Substances Selected

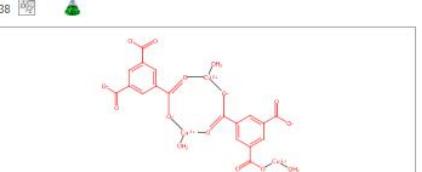
1. 222404-02-6



$C_{18} H_{12} Cu_3 O_{15}$
Copper, triqua[μ -[1,3,5-benzenetricarboxylato(3-)κO:κO¹]]
[μ -[1,3,5-benzenetricarboxylato(3-)κO:κO¹]]tri-

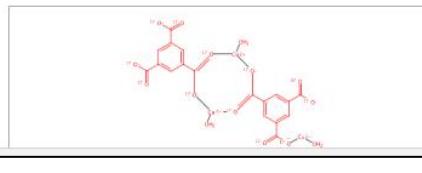
Key Physical Properties
Experimental Properties

2. 222403-98-7

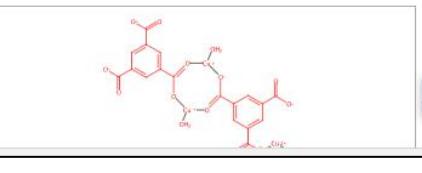


$C_{18} H_{12} Cu_3 O_{15} \cdot x H_2O$
Copper, triqua[μ -[1,3,5-benzenetricarboxylato(3-)κO:κO¹]]
[μ -[1,3,5-benzenetricarboxylato(3-)κO:κO¹]]tri-,
hydrate (1:?)

4. 1685249-53-9



5. 2019181-18-9



SciFinder中的物质记录

Substance Identifier "222404-02-6" > substances (1) > 222404-02-6 > get reactions (222)

SUBSTANCES ? Get References Get Reactions Get Commercial Sources Tools

Analyze Refine

Analyze by: Substance Role

Analytical Study 1

Biological Study 1

Formation, Nonpreparative 1

Occurrence 1

Preparation 1

Process 1

Properties 1

Reactant or Reagent 1

Uses 1

Show More

Sort by: CAS Registry Number

0 of 1 Substance Selected

1. 222404-02-6

~1274

C₁₈H₁₂Cu₃O₁₅ Copper, triquaqua[μ-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻][μ₃-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻:κO²⁻]]tri-

Key Physical Properties Experimental Properties

Chemical Structure

CAS Registry Number: 222404-02-6

- View Substance Detail
- Explore by Structure
- Synthesize this...
- Get Reactions where Substance is a ▾
- Get Commercial Sources
- Get Regulatory Information
- Get References
- Export as Image
- Export as molfile
- Send to SciPlanner

获得其文献、反应和供应商信息

点击CAS号，获得物质信息详情

在SciFinder中，鼠标滑过物质，即可打开物质标准菜单，获得与物质相关的所有内容

SciFinder中的物质记录

Substance Identifier "222404-02-6" > substances (1) > 222404-02-6

SUBSTANCE DETAIL   

 Return

CAS Registry Number 222404-02-6

~1,274  

C₁₈H₁₂Cu₃O₁₅

Copper, triqua[μ-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻][μ₃-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻]]]tricopper

Coordination Compound

Density (Experimental)

Value: 0.879 g/cm³

Other Names

Copper, triqua[μ-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻][μ₃-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻]]]tris-(9CI)

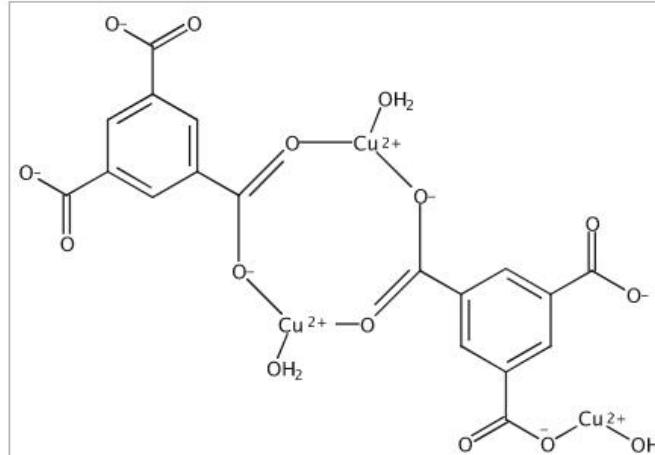
Triqua[μ-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻][μ₃-[1,3,5-benzenetricarboxylato(3-)κO²⁻:κO²⁻]]]tricopper

Basolite C 300

Cu-BTC

Cu₃BTC₂

[View more...](#)



实验数据与实验谱图

EXPERIMENTAL PROPERTIES

Density	Electrical	Mechanical	Nuclear	Optical and Scattering	Structure Related	Thermal	
Density Properties				Value		Condition	Note
Density				0.879 g/cm ³			(5)CAS
Density				See full text		1 of 3	(6)CAS
Notes							
(5) Chen, Banglin; US 20110269984 A1 2011 CAPLUS							
(6) Liu, Ying Dan; Chemical Communications (Cambridge, United Kingdom) 2012, V48(45), P5635-5637 CAPLUS							

EXPERIMENTAL SPECTRA

¹ H NMR	¹³ C NMR	IR	Raman	UV and Visible	X-Ray	Additional Spectra	
¹ H NMR Properties				Value		Condition	Note
Proton NMR Spectrum				See full text			(15)CAS
Notes							
(15) Chen, Joseph J.; Angewandte Chemie, International Edition 2013, V52(46), P12043-12046 CAPLUS							

SciFinder中的物质记录

▼CAS REFERENCE ROLES

Roles	Patents	Nonpatents	Nonspecific Derivatives from Patents	Nonspecific Derivatives from Nonpatents
Analytical Study	✓	✓	✓	✓
Biological Study	✓	✓	✓	✓
Combinatorial Study	✓	✓	✓	
Formation, Nonpreparative	✓	✓	✓	✓
Miscellaneous	✓	✓	✓	✓
Occurrence	✓	✓	✓	✓
Preparation	✓	✓	✓	✓
Process	✓	✓	✓	✓
Properties		✓	✓	
Prophetic in Patents			✓	
Reactant or Reagent			✓	✓
Uses	✓	✓	✓	✓

选中任何一项即可获得相关文献

如何使用SciFinder提高科研效率

- 以“光解水”为例，如何利用SciFinder：

- 获取研究全景
 - 对文献结果进行分析、筛选、及精准定位所需信息

- 物质检索

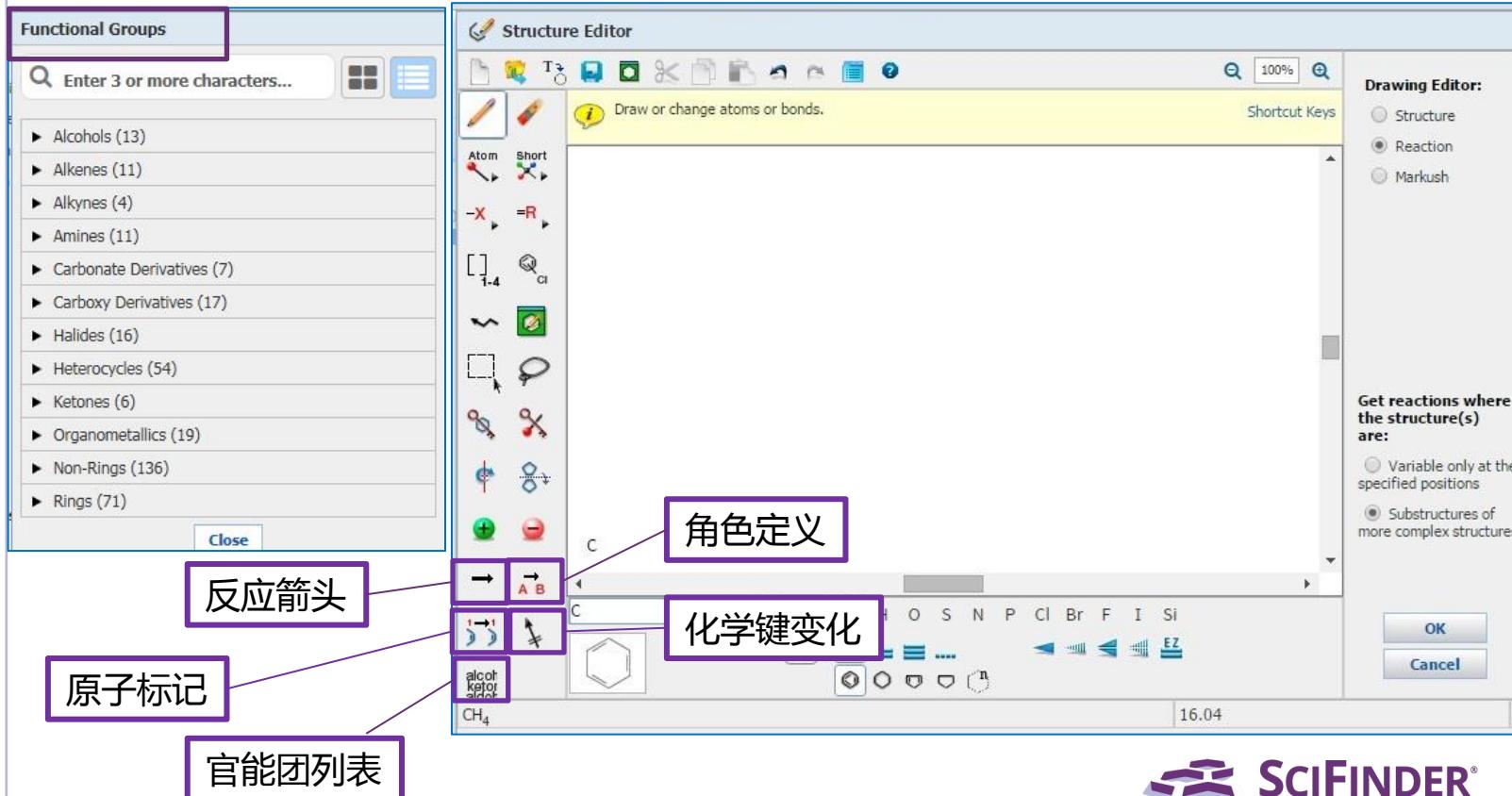
- 如何从大量物质结果集中快速定位目标物质
 - 结构式检索；分子式检索（合金，多氧簇金属，MOF化合物）；物质属性检索
 - 如何获得物质的属性信息、谱图信息
 - 化合物分析方法检索

- 反应检索

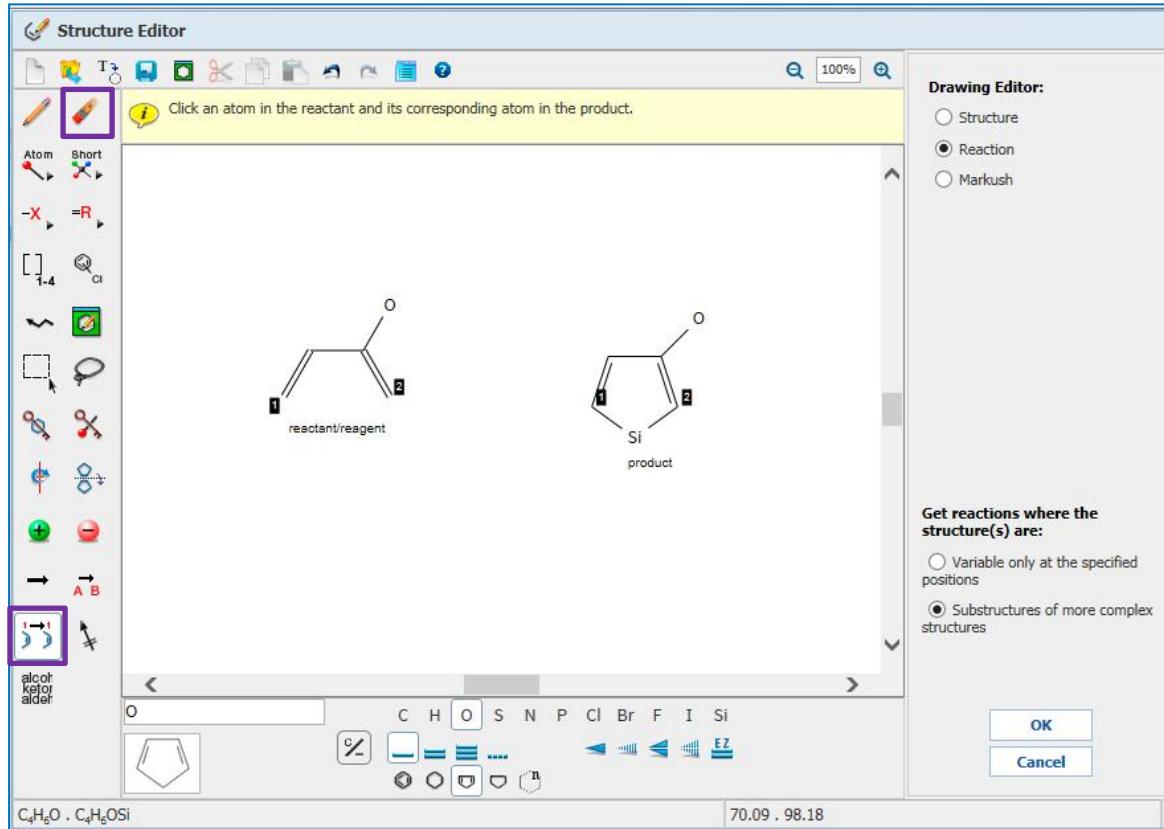
- 结构编辑器的使用（绘图、限定反应的工具）
 - 如何进行精确结构反应检索，并获得实验详情
 - 如何进行亚结构反应检索，以及如何处理检索结果
 - 案例分享
 - 多氧簇金属的反应信息
 - MOF化合物反应信息



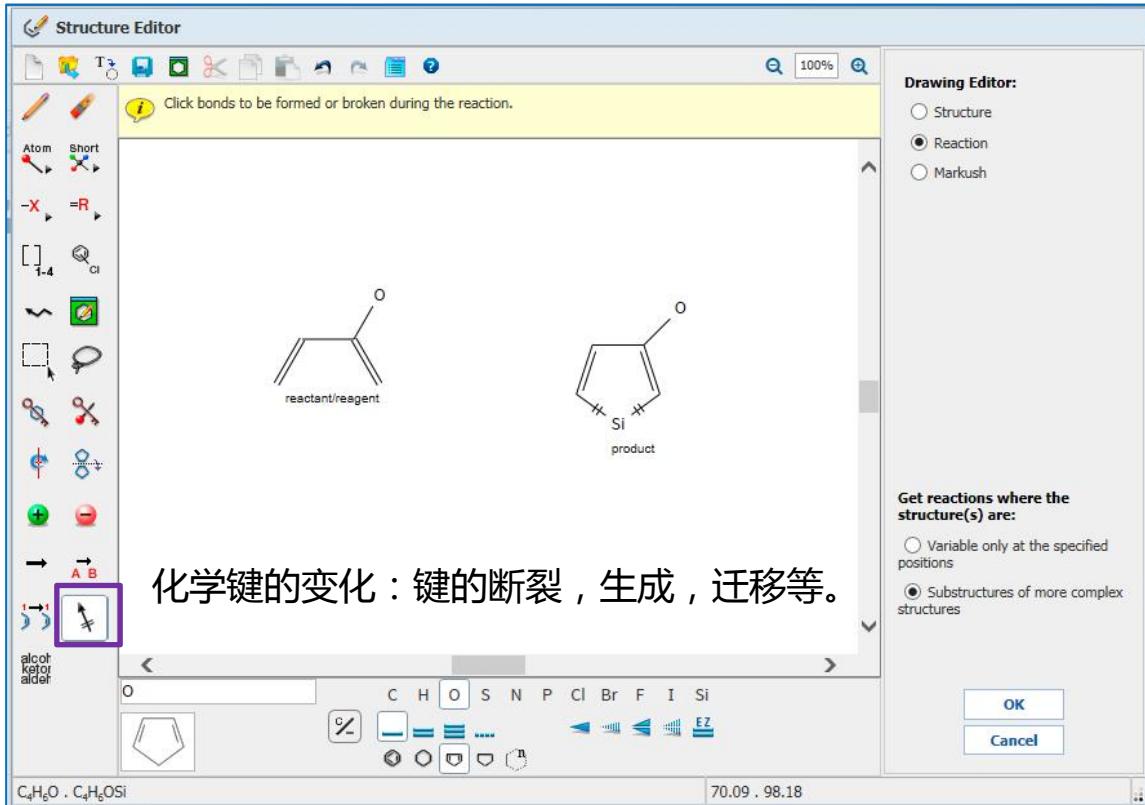
结构编辑器：绘制反应工具



例1：反应前后物质中的原子标记

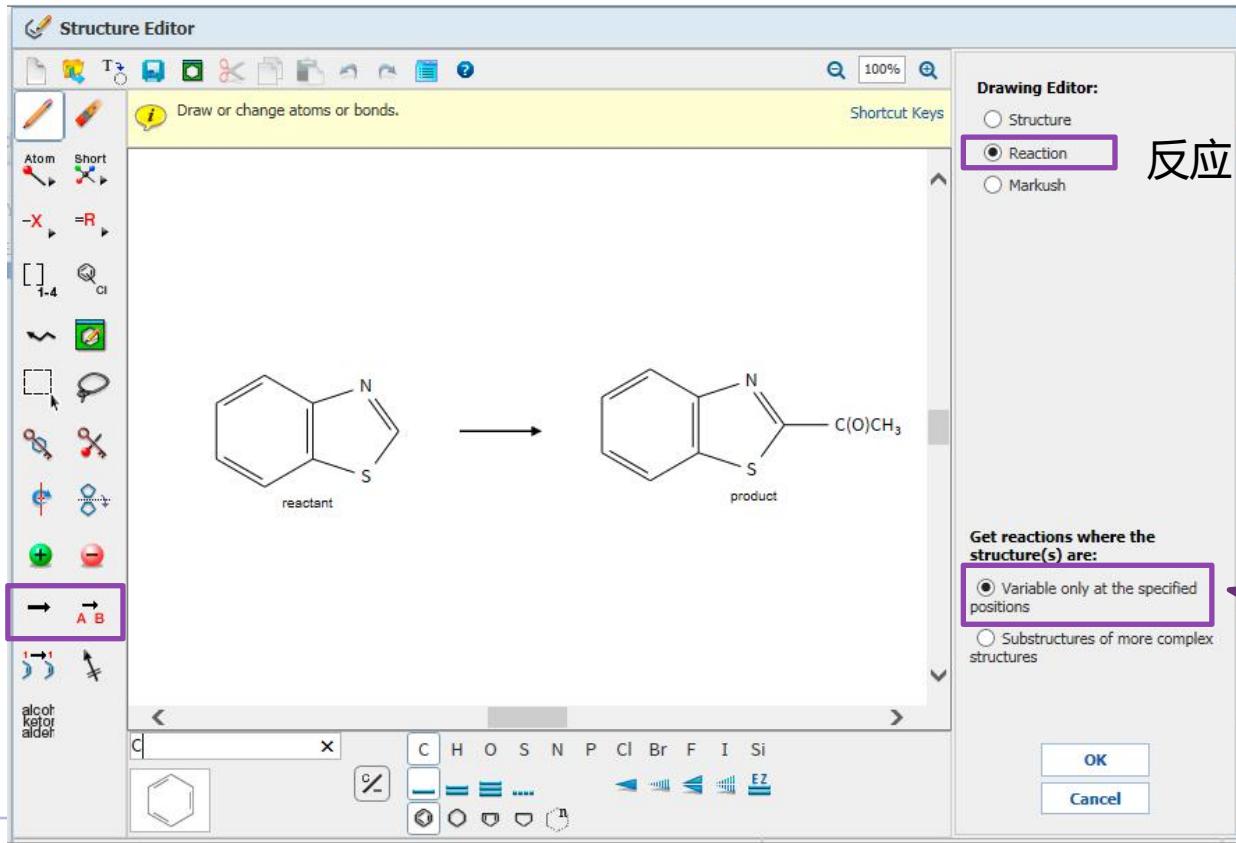


例2：发生变化的化学键标记



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精确结构反应检索：绘制反应式



执行的是锁环
锁原子的检索

NDER®

精确结构反应检索：查看反应结果集

Reaction Structure structure variable only at spe... > reactions (8)

REACTIONS ? Get References Tools ▾

Analyze Refine Group by: No Grouping Sort by: Number of Steps ↑

0 of 8 Reactions Selected

1. View Reaction Detail Similar Reactions

Single Step Hover over any structure for more options.

Analyze by: Reagent

Reagent	Count
BuLi	4
t-BuOOH	3
HCl	2
19468-88-3	1
H ₂ O	1
H ₂ SO ₄	1

Show More

Chemical structures:

Reaction 1: 2-thiophenyl ring + methyl isocyanide ($\text{CH}_3\text{N}=\text{C}(=\text{O})\text{CH}_3$)

Reaction 2: 2-thiophenyl ring + methyl isocyanide ($\text{CH}_3\text{N}=\text{C}(=\text{O})\text{CH}_3$)

Overview: 1.1 R:BuLi, S:THF, S:Me(CH₃) -78°C; 1 h, -78°C

Export

Export: All Selected Range Example: 2-20

For: Offline review Portable Document Format (*.pdf) Rich Text Format (*.rtf)

Saving locally Answer Key eXchange (*.akex)

Details: File Name: * Reaction_05_05_2017_171255.x

Format: Summary Detail

Include: Experimental Procedure (if available) MethodsNow™ Protocols (if available) Overview Task History

Export Cancel

分组，排序 点击Export导出结果

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Copyright 2016 American Chemical Society reserved.

精确结构反应检索：查看反应结果集

点击Document，合并来自同一篇文献的反应；
点击Transformation，获得反应类型的分类。

Reaction Structure structure variable only at spe... > reactions (8)

REACTIONS ?

Analyze Refine

Analyze by: Reagent

BuLi 4
*t*BuOOH 3
HCl 2
19468-88-3 1
H₂O 1
H₂SO₄ 1

Show More

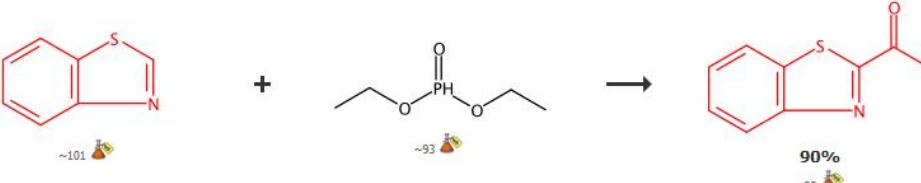
Get References Tools ▾

Group by: No Grouping Document Transformation 8 of 8 Reactions Selected

Sort by: Relevance

1. View Reaction Detail

Single Step Hover over any structure for more options.



Overview

Steps/Stages

1.1 R:*t*BuOOH, S:Me(CH₂)₂Me, 24 h, 80°C
1.2 R:H₂O

Notes

optimized on amount of TBHP, amount of phosphate, temperature and solvent, using hydrogen peroxide in MeCN resulted in lower yield, optimization study, Reactants: 2, Reagents: 2, Solvents: 1, Steps: 1, Stages: 2, Most stages in any one step: 2

References

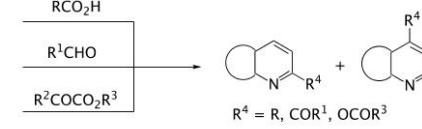
Peroxides as "Switches" of Dialkyl H-Phosphonate: Two Mild and Metal-Free Methods for

Get References Tools ▾

Group by: Transformation Sort by: Frequency

0 of 8 Reactions Selected

1. Decarboxylative Alkylation, Acylation and Carbalkoxylation of Nitrogen Heterocycle 5 Reactions



2. Uncategorized Single-Step Reactions 2 Reactions

3. Multi-Step Reactions 1 Reaction

精确结构反应检索：查看反应结果集

排序：相关度，入库号，实验步骤，MethodsNow, 步数，产率，发表年份

Reaction Structure structure variable only at spe... > reactions (8)

REACTIONS

Analyze Refine

Analyze by: Reagent

Reagent	Count
BuLi	4
t-BuOOH	3
HCl	2
19468-88-3	1
H ₂ O	1
H ₂ SO ₄	1

Get References Tools

Group by: No Grouping Sort by: Number of Steps

0 of 8 Reactions Selected

1. View Reaction Detail Link Similar Reactions

Single Step Hover over any structure for more options.

2-thiophenylmagnesium bromide + 2-bromo-N,N-dimethylacetamide → 2-acetyl-3-thiophenylmethyl acetate (58%)

~85 °C ~102 °C ~79 °C

Overview

Steps/Stages

1.1 R:BuLi, S:THF, S:Me(CH₂)₄Me, -78°C; 1 h, -78°C

Notes

ice-bath removed after stirring at -78°C for 1 hour (stage 2), Reactants: 2, Reagents: 2,

精确结构反应检索：查看反应结果集

Reaction Structure structure variable only at spe... > reactions (8)

REACTIONS Get References Tools ▾ Send to SciPlanner Display Options

Analyze Refine

Analyze by: Reagent

Reagent	Count
BuLi	4
t-BuOOH	3
HCl	2
19468-88-3	1
H ₂ O	1
H ₂ SO ₄	1

Group by: No Grouping Document Transformation Sort by: Relevance

1. View Reaction Detail Link Similar Reactions

Single Step Hover over any structure for more options.

对于一步反应，可以点击 similar Reaction, 获取相似反应

Reaction Overview:

Reactants:

- 1. R: t-BuOOH, S: Me(CH₂)₂Me, 24 h, 80°C
- 2. R: H₂O

Yield: 90%

Notes:

optimized on amount of TBHP, amount of phosphate, temperature and solvent, using hydrogen peroxide in MeCN resulted in lower yield, optimization study, Reactants: 2, Reagents: 2, Solvents: 1, Steps: 1, Stages: 2, Most stages in any one step: 2

References:

Peroxides as "Switches" of Dialkyl H-Phosphonate: Two Mild and Metal-Free Methods for

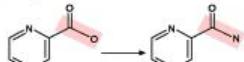
精确结构反应检索：获取相似反应

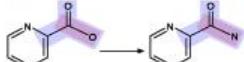
Get Similar Reactions [?](#)

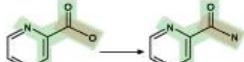
Retrieve similar reactions from:

All reactions
 Current answer set

Include this level of similarity:

Broad - Reaction centers only


Medium - Reaction centers plus adjacent atoms and bonds


Narrow - Reaction centers plus extended atoms and bonds


[Get Reactions](#) [Cancel](#)

相似度限制：

Broad : 仅反应中心相似

Medium : 反应中心及附属原子和键

Narrow : 反应中心及扩展的原子和键

精确结构反应检索：查看感兴趣的反应信息

0 of 8 Reactions Selected

1. View Reaction Detail Similar Reactions

Single Step Hover over any structure for more options.

~101
+ ~93
→ 90%

▼ Overview

Steps/Stages

1.1 R:*t*-BuOOH, S:Me(CH₂)₅Me, 24 h, 80°C
1.2 R:H₂O

Notes

optimized on amount of TBHP, amount of phosphate, temperature and solvent, using hydrogen peroxide in MeCN resulted in lower yield, optimization study, Reactants: 2, Reagents: 2, Solvents: 1, Steps: 1, Stages: 2, Most stages in any one step: 2

References

Peroxides as "Switches" of Dialkyl H-Phosphonate: Two Mild and Metal-Free Methods for Preparation of 2-Acylbenzothiazoles and Dialkyl Benzothiazol-2-ylphosphonates
Quick View Other Sources
By Chen, Xiao-Lan et al
From Journal of Organic Chemistry, 79(17), 8407-8416; 2014

实验步骤

▼ Experimental Procedure

JOC
The Journal of Organic Chemistry

Experimental Procedures for the Synthesis of 2-Acylbenzothiazoles (3a-3ab) A mixture of benzothiazole (135.0 mg, 1.0 mmol), phosphonate (5.0 mmol), and TBHP (10.0 mmol) in CH₃CN (2.0 mL) was stirred at 80 °C for 24 h. The reaction mixture was quenched with water (5.0 mL) and extracted with ethyl acetate (3 × 5.0 mL). The combined organic layers were washed with brine (15.0 mL) and dried over anhydrous MgSO₄. After filtration, the solvent was evaporated in vacuo. The crude product was purified by silica gel chromatography (petroleum ether/ethyl acetate 20/1) to give the desired product **1-(Benzothiazol-2-yl)ethanone (3a)**: yield 90%. mp 107–110 °C; ¹H NMR (400 MHz, CDCl₃) δ 2.83 (s, 3H), 7.53 (td, *J* = 7.6, 1.3 Hz, 1H), 7.58 (td, *J* = 8.0, 1.3 Hz, 1H), 7.98 (*d*, *J* = 8.0 Hz, 1H), 8.18 (*d*, *J* = 7.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 26.2, 122.5, 125.5, 127.0, 127.7, 137.5, 153.6, 166.5, 193.2; HRMS (ESI) calcd for C₉H₁₀NOS [M + H]⁺, 178.0321, found 178.0320.

67

MethodsNow Synthesis——人工标引的反应信息，节省您宝贵的时间

- 详细、明确的物质信息
- 全面、有条理的实验过程信息
- 更好的阅读体验——表格形式
- 无需查看原文直接获取实验详情——反应物，反应条件，步骤，产物性质，谱图等

MethodsNow Synthesis: 通过Analyze, 选择有MethodsNow标引的反应，点击MethodsNow查看实验详情

□ 3. View Reaction Detail Similar Reactions

Single Step Hover over any structure for more options.

~101
+
58%

▼ Overview

Steps/Stages

1.1 R:BuLi, S:THF, S:Me(CH₂)₂Me, -78°C; 1 h, -78°C
1.2 1 h, -78°C; 10 min
1.3 R:HCl, S:H₂O, rt, acidify

Notes

ice-bath removed after stirring at -78C for 1 hour (stage 2), Reactants: 2, Reagents: 2, Solvents: 3, Steps: 1, Stages: 3, Most stages in any one step: 3

References

Rhodium Catalyzed Asymmetric Hydrogenation of 2-Pyridine Ketones

By Yang, Hailong et al
From Organic Letters, 17(17), 4144-4147; 2015

▼ METHODS NOW™

Procedure

1. Cool the solution (0.5 M) of benzothiazole in dry THF under nitrogen to -78 °C.
2. Add dropwise nBuLi (1.1 equiv, 5.5 mmol, 2.3M in hexane).

[View more...](#)

Available Experimental Data

¹H NMR, ¹³C NMR, State

[View with MethodsNow](#)

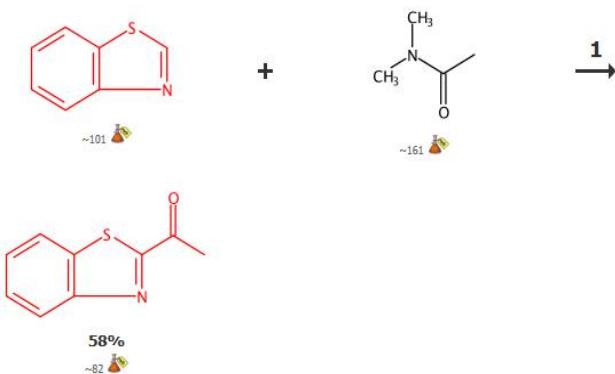
点击查看MethodsNow窗口

MethodsNow Synthesis: 实验详情展示窗口

MethodsNow

Rhodium Catalyzed Asymmetric Hydrogenation of 2-Pyridine Ketones

By Yang, Hailong; Huo, Ningning; Yang, Ping; Pei, Hao; Lv, Hui; Zhang, Xumu
From Organic Letters, 17(17), 4144-4147; 2015
Published by American Chemical Society



产物，反应物，试剂，溶剂，步骤，
反应类型，规模，核磁氢谱，核磁碳谱，
产物状态，CAS方法号

Products

Ethanone, 1-(2-benzothiazolyl)-, 58%, CAS RN: 1629-78-3

Reactants

Benzothiazole, CAS RN: 95-16-9
Dimethylacetamide, CAS RN: 127-19-5

Reagents

Butyllithium, CAS RN: 109-72-8
Hydrochloric acid, CAS RN: 7647-01-0

Solvents

Tetrahydrofuran, CAS RN: 109-99-9
Hexane, CAS RN: 110-54-3
Water, CAS RN: 7732-18-5

Procedure

- Cool the solution (0.5 M) of benzothiazole in dry THF under nitrogen to -78 °C.
- Add dropwise nBuLi (1.1 equiv, 5.5 mmol, 2.3M in hexane).
- Keep the resulted mixture for 1 h under -78 °C.
- Add N,N-dimethylacetamide (1 equiv, 5 mmol) to the stirred solution at -78 °C.
- Stir the mixture was continuously for 1 h.
- Remove the cold bath and stir the mixture for additional 10 min.
- Hydrolyze with concd hydrochloric acid (1 mL).
- Stir the acidic solution continuously until the temperature reach to room temperature.
- Pour the mixture into same amount of water.
- Extract the aqueous mixture with ethyl acetate 3 times.
- Dry the combined organic solution with Na₂SO₄.
- Purify by flash column chromatography to afford product.

Transformation

Decarboxylative Alkylation, Acylation and Carbalkoxylation of Nitrogen Heterocycle

¹H NMR

(400 MHz, CDCl₃)δ8.22 (d,J = 8.0Hz,1H), 8.01 (d, J = 7.6Hz, H), 7.55-7.63 (m, 2H), 2.87 (s, 3H);

¹³C NMR

(101MHz, CDCl₃)δ193.2, 166.5, 153.6, 137.4, 127.7, 127.0, 125.5, 122.5, 26.2,

State

white solid

CAS Method Number

3-219-CAS-1662290

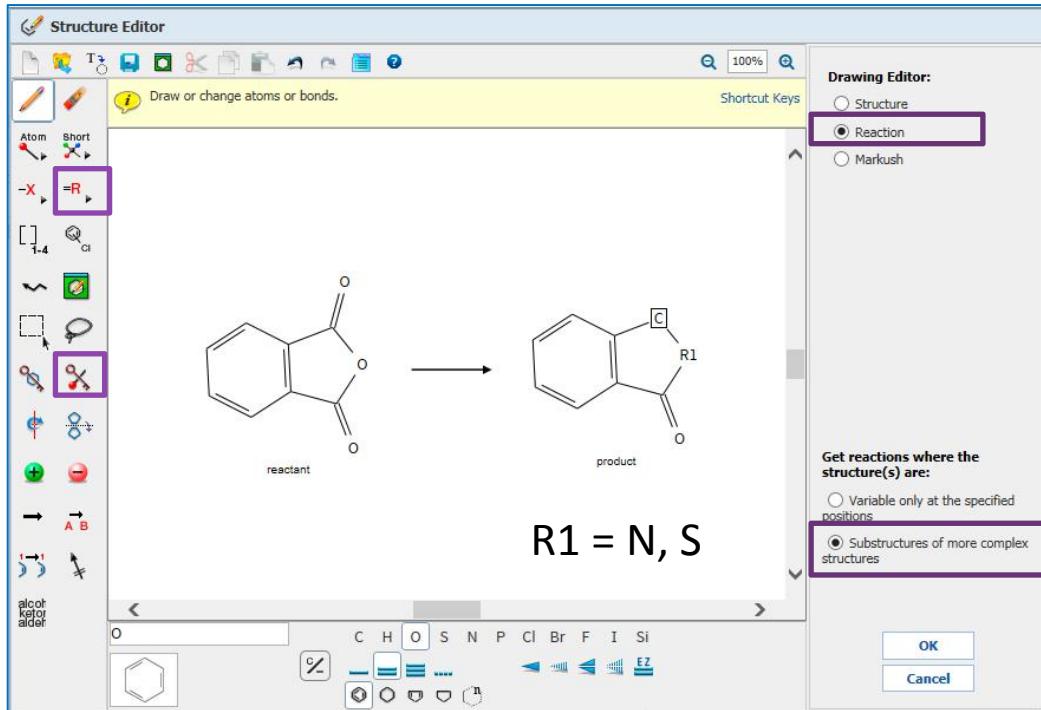
PDF or XLS格式

Print/Export Close

提纲

- SciFinder中获取化合物制备信息的方法
- 物质相关信息获取方法
 - 结构编辑器的使用（绘图工具，定义和限定化合物等）
 - 如何进行物质检索（结构，属性，识别号检索等）
 - 如何查看物质详情
- 反应相关信息获取方法
 - 结构编辑器的使用（绘图、限定反应的工具）
 - 如何进行精确结构反应检索，并获得实验详情
 - 如何进行亚结构反应检索，以及如何处理检索结果
 - 反应检索案例分享

亚结构反应检索: 绘制反应式



输入的反应物和产物
结构可以被修饰，但
母体结构不变

亚结构反应检索：获得反应结果集

REACTIONS ?

Analyze Refine

Refine by: Reaction Structure

Product Yield Number of Steps Reaction Classification Excluding Reaction Classification Non-participating functional groups

Structure Editor: Java Non-Java

Click image to change structure or view detail.

Search type: Substructure

Refine

Get References Tools ▾

Group by: No Grouping Sort by: Relevance ↴

Display Options

Send to SciPlanner

0 of 1208 Reactions Selected

Page: 1 of 81

1. **View Reaction Detail**

2 Steps Hover over any structure for more options.

Overview

Steps/Stages

1.1 R:H₂NCHO
2.1 R:HCl, R:Sn, S:H₂O, S:EtOH

Notes

Reactants: 1, Reagents: 3, Solvents: 2, Steps: 2, Stages: 2, Most stages in any one step: 1

References

Studies on the chemistry of isoindoles and isoindolenines. XXVII. 3-Alkoxy-1H-isoindoles: syntheses and properties

Quick View Other Sources

By Hennige, Hans et al
From *Chemische Berichte*, 121(2), 243-52; 1988

亚结构反应检索：分析处理

13种分析选项

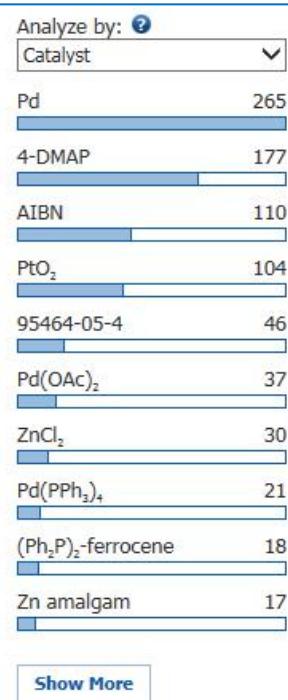
Analyze Refine

Analyze by: ?

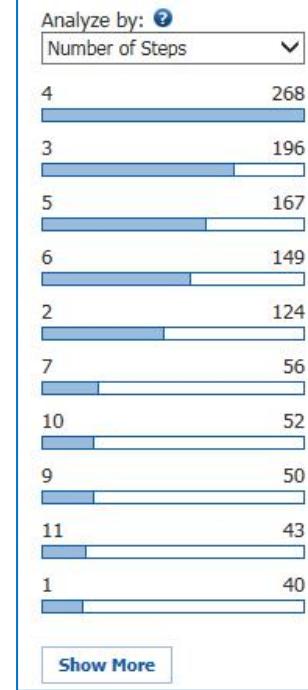
- Author Name
- Catalyst
- Company-Organization
- Document Type
- Experimental Procedure
- Journal Name
- Language
- MethodsNow
- Number of Steps
- Product Yield
- Publication Year
- Reagent
- Solvent



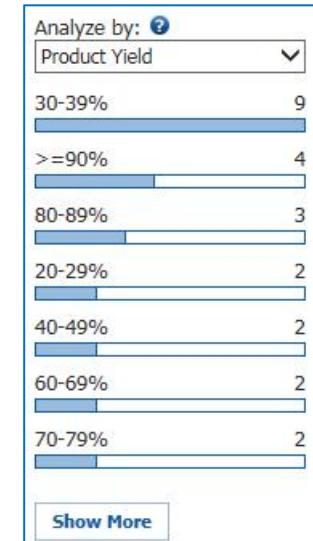
催化剂



反应步数

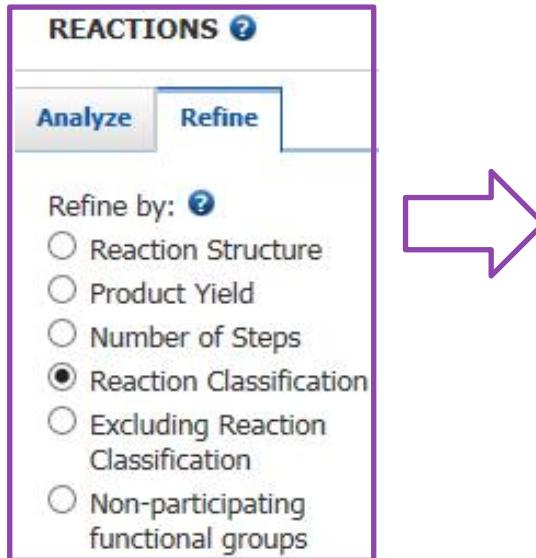


产率



亚结构反应检索：筛选处理

6种筛选选项：反应结构、产率、反应步数、包含/排除的反应类型(11种)、反应官能团(217种)



- Reaction Classification(s):
- Biotransformation
 - Catalyzed
 - Chemoselective
 - Combinatorial
 - Electrochemical
 - Gas-phase
 - Non-catalyzed
 - Photochemical
 - Radiochemical
 - Regioselective
 - Stereoselective
- Refine

亚结构反应检索：勾选反应类型

REACTIONS ② Get References Tools ▾ Send to SciPlanner

Analyze Refine Group by: No Grouping Sort by: Relevance Display Options

Refine by: ?
 Reaction Structure
 Product Yield
 Number of Steps
 Reaction Classification
 Excluding Reaction Classification
 Non-participating functional groups

0 of 1217 Reactions Selected

Page: 1 of 82

1. View Reaction Detail

2 Steps Hover over any structure for more options.

Reaction Classification(s):
 Biotransformation
 Catalyzed
 Chemoselective
 Combinatorial
 Electrochemical
 Gas-phase
 Non-catalyzed
 Photochemical
 Radiochemical
 Regioselective
 Stereoselective

Refine

Overview

Steps/Stages

1.1
2.1

Notes

1) no experimental details, prophetic reaction, 2) literature preparation, prophetic reaction, no experimental details, Reactants: 1, Steps: 2, Stages: 2, Most stages in any one step: 1

References

Lenalidomide isotopologues and their preparation and use for the treatment of diseases

Quick View PATENTPAK ▾
By Muller, George W. and Man, Hon-Wah
From PCT Int. Appl., 2010093434, 19 Aug 2010

亚结构反应检索：排除反应类型

Reaction Structure substructure > reactions (1208) > refine "Electrochemical Gas-phase Phot..." (1154)

REACTIONS

Analyze Refine Display Options

Refine by: Reaction Structure Product Yield Number of Steps Reaction Classification Excluding Reaction Classification Non-participating functional groups

Excluding Reaction Classification(s): Biotransformation Catalyzed Chemoselective Combinatorial Electrochemical Gas-phase Non-catalyzed Photochemical Radiochemical Regioselective Stereoselective

0 of 1154 Reactions Selected Page: 1 of 77

1. [View Reaction Detail](#)

2 Steps Hover over any structure for more options.

Overview

Steps/Stages

1.1 R:H₂NCHO
2.1 R:HCl, R:Sn, S:H₂O, S:EtOH

Notes

Reactants: 1, Reagents: 3, Solvents: 2, Steps: 2, Stages: 2, Most stages in any one step: 1

References

Studies on the chemistry of isoindoles and isoindolenines. XXVII. 3-Alkoxy-1H-isoindoles: syntheses and properties

By Henning Hanc et al.

亚结构反应检索：筛选官能团

Reaction Structure substructure > reactions (1208) > refine "Electrochemical Gas-phase Phot..." (1154) > refine "any HETEROCYCLES KETONES" (435)

REFINERIES [Get References](#) [Tools](#) [Send to SciPlanner](#)

Analyze Refine Group by: No Grouping Sort by: Relevance [Display Options](#)

Refine by: Reaction Structure Product Yield Number of Steps Reaction Classification Excluding Reaction Classification Non-participating functional groups

Non-participating Functional Group(s) View: Classes 10

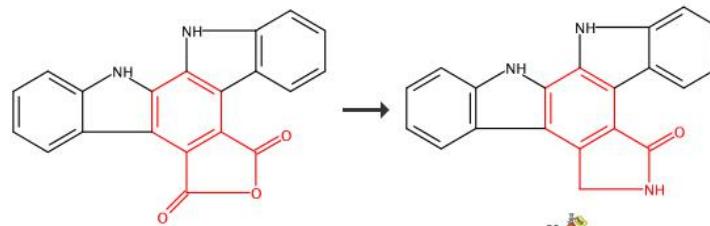
2 Selected [Clear Selections](#)

AMINES CARBONATE DERIVATIVES CARBOXY DERIVATIVES HALIDES HETEROCYCLES KETONES ORGANOMETALLICS

0 of 435 Reactions Selected Page: 1 of 29

1. [View Reaction Detail](#) [Link](#)

2 Steps Hover over any structure for more options.



Overview

Steps/Stages

1.1 R:NH₂OAc, 3 h, 140°C
2.1

Notes

1) thermal, 2) literature preparation, no experimental detail, Reactants: 1, Reagents: 1, Steps: 2, Stages: 2, Most stages in any one step: 1

References

查找多氧簇金属的反应信息：

CAS Registry Number 76210-25-8

-612
C₃₂ H₅₆ Mn₁₂ O₄₈

也可以获得反应信息

Manganese, hexadecakis[μ-(acetato- κO)]tetraaquadodeca- μ_3 -oxododeca-, compd. with acetic acid, hydrate (1:2:4)

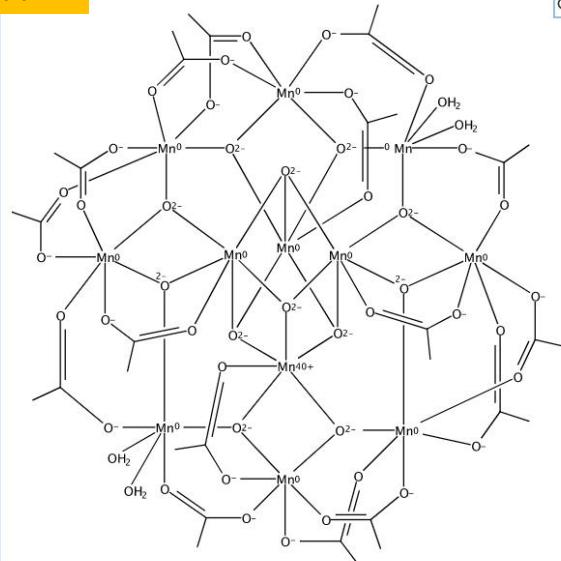
Other Names

Manganese, hexadecakis[μ-(acetato- κO)]tetraaquadodeca- μ_3 -oxododeca-, compd. with acetic acid (1:2), tetrahydrate (9CI)
Manganese, hexadecakis[μ-(acetato- κO)]tetraaquadodeca- μ_3 -oxododeca-, compd. with acetic acid (1:2), tetrahydrate (9CI)
Hexadecaacetatotetraquadecaoxododecamanganese-acetic acid tetrahydrate (1:2)

Mn₁₂

Mn₁₂ acetate

[View more...](#)



64-19-7
C₂H₄O₂



CAS Registry Number: 76125-82-1

[View Substance Detail](#)

[Explore by Structure](#)

[Synthesize this...](#)

[Get Reactions where Substance is a](#)

Product

Reactant

Reagent

Reactant/Reagent

Catalyst

Solvent

Any Role

查看反应信息详情

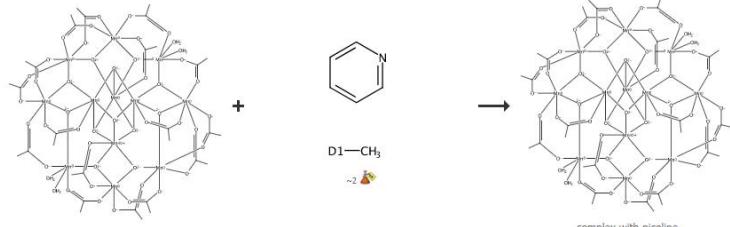
Analyze Refine Group by: No Grouping Sort by: Accession Number Display Options

Analyze by: Reagent
KMnO₄ 6
TEA chloride 2
110743-42-5 1
Ce(NH₄)₂(NO₃)₆ 1
H₂O 1
Mn(OAc)₂ 1
[Show More](#)

0 of 12 Reactions Selected

1. View Reaction Detail [Link](#)

Single Step Hover over any structure for more options.



complex with picoline

Overview

Steps/Stages

1.1 R:Ce(NH₄)₂(NO₃)₆, S:MeCN, rt; 15 min, rt

Notes

crystals formed after 3 days at room temperature, Reactants: 2, Reagents: 1, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

References

Manganese/Cerium Clusters Spanning a Range of Oxidation Levels and CeMn8, Ce2Mn4, and Ce6Mn4 Nuclearities: Structural, Magnetic, and EPR Properties
Quick View Other Sources
By Lampropoulou, Christos et al
From Inorganic Chemistry, 53(13), 6805-6816; 2014

Experimental Procedure

Inorganic Chemistry General/Typical Procedure: **Method B.** To a stirred solution of [Mn₁₂O₁₂(O₂CMe)₁₆(H₂O)₄] (0.50 g, 0.25 mmol) in a solvent mixture comprising MeCN (15 mL) and pyridine (10 mL) was slowly added solid (NH₄)₂[Ce(NO₃)₆] (0.28 g, 0.50 mmol). The resulting solution was stirred for 15 min, during which time the color changed slightly from dark brown to reddish brown. The solution was filtered and left undisturbed for a period of 3 d, during which time black needles of **1**·2py·6MeCN grew. They were isolated as for Method A; the yield was based on Ce. The identity of the product was confirmed by elemental analysis, IR spectral comparison, and unit cell determination to be identical to material from Method A. The picoline version [Ce₆Mn₄O₁₂(O₂CMe)₁₆(NO₃)₄][pic]₂] (**1'**) can be obtained in comparable yield by the same method using picoline instead of pyridine. Anal. Calcd (Found) for **1'**·2H₂O: C, 21.14 (21.25); H, 2.50 (2.75); N, 4.48 (4.26)%.

METHODSNOW™

Procedure

- Slowly add solid (NH₄)₂[Ce(NO₃)₆] (0.28 g, 0.50 mmol) to a stirred solution of [Mn₁₂O₁₂(O₂CMe)₁₆(H₂O)₄] (0.50 g, 0.25 mmol) in a solvent mixture comprising MeCN (15 mL) and picoline (10 mL).

[View more...](#)

Available Experimental Data
Elemental Analysis
[View with MethodsNow](#)

查找MOF的反应信息：

Substance Identifier "222404-02-6" > substances (1) > 222404-02-6

SUBSTANCE DETAIL [?](#) [Get References](#) [Get Reactions](#)

[Return](#)

CAS Registry Number 222404-02-6

~1,274

C₁₈H₁₂Cu₃O₁₅
Copper, triqua[μ -[1,3,5-benzenetricarboxylato(3-)- κ^O : κ^O]][μ ₃-[1,3,5-benzenetricarboxylato(3-)- κ^O : κ^O : κ^O]]tris-

Coordination Compound

Density (Experimental)
Value: 0.879 g/cm³

Other Names
Copper, triqua[μ -[1,3,5-benzenetricarboxylato(3-)- κ^O : κ^O]][μ ₃-[1,3,5-benzenetricarboxylato(3-)- κ^O : κ^O : κ^O]]tris-(9Cl)
Triqua[μ -[1,3,5-benzenetricarboxylato(3-)- κ^O : κ^O : κ^O]][μ ₃-[1,3,5-benzenetricarboxylato(3-)- κ^O : κ^O : κ^O]]tricopper
Basolite C 300
Cu-BTC
Cu₃BTC₂
[View more...](#)

CAS Registry Number: 222404-02-6

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[Get Reactions where Substance is a](#) [Product](#)

[Get Commercial Sources](#) [Reactant](#)

[Get Regulatory Information](#) [Reagent](#)

[Get References](#) [Reactant/Reagent](#)

[Export as Image](#) [Catalyst](#)

[Export as molfile](#) [Solvent](#)

[Send to SciPlanner](#) [Any Role](#)

EXPERIMENTAL PROPERTIES

EXPERIMENTAL SPECTRA

CAS REFERENCE ROLES

ADDITIONAL DETAILS

查看反应信息详情

Substance Identifier "222404-02-6" > substances (1) > 222404-02-6 > get reactions (222) > keep analysis "Experimental Procedure" (198)

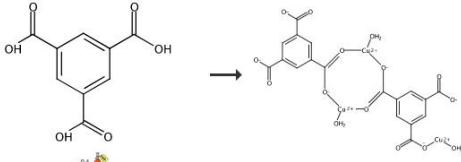
REACTIONS  Get References Tools 

Analyze Refine Analyze by: Solvent Group by: No Grouping Sort by: Accession Number  Display Options  Page: 1 of 2 

0 of 198 Reactions Selected

3. View Reaction Detail  

Single Step Hover over any structure for more options.



Overview

Steps/Stages

1.1 R: $\text{Cu}(\text{NO}_3)_2$, S: H_2O , S: MeOH , S: DMF , 10 min, 90°C

Notes

microwave irradiation, microwave synthesizer (500W) used, Reactants: 1, Reagents: 1, Solvents: 3, Steps: 1, Stages: 1, Most stages in any one step: 1

References

Metal-Organic Frameworks with Incorporated Carbon Nanotubes: Improving Carbon Dioxide and Methane Storage Capacities by Lithium Doping
 Quick View  Other Sources
By Xiang, Zhonghua et al.
From Angewandte Chemie, International Edition, 50(2), 491-494, S491-1-S491/17; 2011

METHODSNow™

Procedure

- Sonicate the benzene-1,3,5-tricarboxylic acid (2.5 g) and copper nitrate hemihexahydrate (5 g) for 15 min in a 125 mL solvent consisting of equal parts of N,N-dimethylformamide (DMF), ethanol, and deionized water in a 300 mL wide mouth glass jar.
- Place the jar inside laboratory microwave synthesizer (XH-MC-1, Xianghu Technology Co., Ltd.).

[View more...](#)

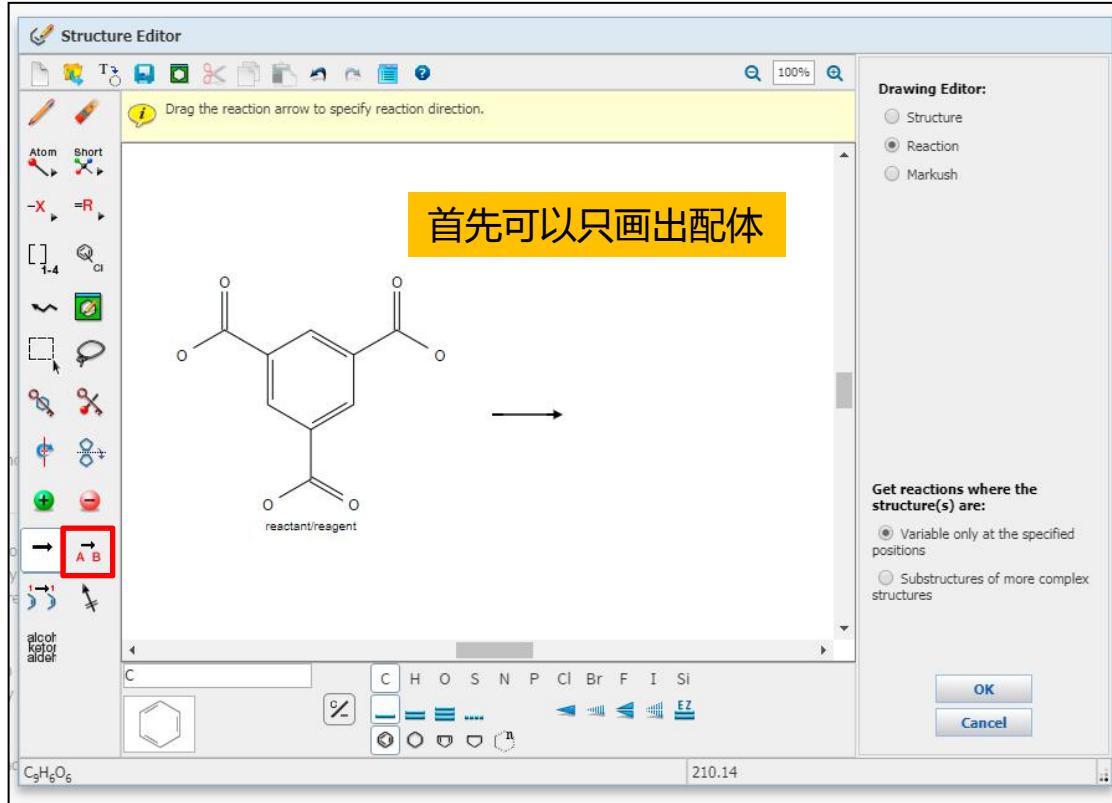
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已知配体查找MOF的合成信息



首先可以只画出配体

在反应结果集中选择Analyze by Reagent, 找到相应金属盐

Reaction Structure structure variable only at spe... > reactions (7218)

REACTIONS 2

Analyze Refine

Analyze by: Reagent

Et₃N 2942
F₃CCO₂H 2596
EtN(Pr-i)₂ 2542
132705-51-2 2111
Na ascorbate 2034
CuSO₄ 2024
K₂CO₃ 1941
NaHCO₃ 1910
NaN₃ 1902
H₂O 1694

Show More

Get References Tools

Group by: No Grouping Sort by: Relevance

0 of 7218 Reactions Selected

1. View Reaction Detail Link

Single Step Hover over any structure for more options.

100%

~84

2. View Reaction Detail Link

Analyze - Reagent

682 Items 2 Selected Export

Sort by: Natural Order Page: 8 of 14

Select bars to view only those reactions within the current answer set.

Reagent	Count
Cu	426
<input checked="" type="checkbox"/> Cu acetylacetone	1
<input checked="" type="checkbox"/> Cu(BF ₃) ₂	1
<input type="checkbox"/> Cu(CF ₃ SO ₃) ₂	1
<input type="checkbox"/> Cu(ClO ₄) ₂	15
<input type="checkbox"/> Cu(NO ₃) ₂	267
<input type="checkbox"/> Cu(NO ₃) ₂ • 3H ₂ O	52
<input type="checkbox"/> Cu(OAc) ₂	73
<input type="checkbox"/> Cu(OAc) ₂ • H ₂ O	20
<input type="checkbox"/> Cu(OH) ₂	12

Apply Cancel

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A CAS SOLUTION

获得相应反应信息

Analyze Refine

Analyze by: Reagent

Group by: No Grouping Sort by: Relevance

0 of 7218 Reactions Selected

1. View Reaction Detail [Link](#)

Single Step Hover over any structure for more options.

Et₃N 2942
F₃CCO₂H 2596
EtN(Pr-*i*)₂ 2542
132705-51-2 2111
Na ascorbate 2034
CuSO₄ 2024
K₂CO₃ 1941
NaHCO₃ 1910
NaN₃ 1902
H₂O 1694

Show More

100%

1.1 R:CuCl₂, R:Propylene oxide, R:

S:H₂O, 5 min

2. View Reaction Detail [Link](#)

Single Step Hover over any structure for more options.

100%

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